

I 23280-66 UNIT(1)  
ACC NR: AP6011437

SOURCE CODE: UR/0020/66/167/004/0925/0927

AUTHORS: Kozlov, V. A.; Saksonov, P. P.; Dobrov, N. N.; Antipov, V. V.;  
Parehin, V. S.

**ORG:** none

TITLE: Altered resistance of animals exposed to vibration<sup>2</sup> to the action  
of some chemical preparations and physical load

SOURCE: AN SSSR, Doklady, v. 167, no. 4, 1966, 925-927

TOPIC TAGS: vibration, cystamine, strychnine, radiation protection, combined stress

**ABSTRACT:** Two series of experiments were conducted on 449 white mice weighing 20-24 g. In the first series, 240 mice were exposed to vibration (70 cps, 0.4 mm, 10 G, 1 hr exposure), after which they were given IP injections of cystamine chlorhydrate (400 mg/kg) or strychnine (1.5 mg/kg) 20 min or 4 hr later. These preparations were selected because they have a therapeutic effect for radiation sickness or injuries and may be used on prolonged spaceflights, should severe radiation conditions occur. It was established that the toxic action of these drugs was elevated in vibrated animals. In the control group, mortality was 45% for cystamine and 47% for strychnine. In the vibrated

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ACC NR: AP6011437

group, these values increased to 53.7% and 61.2% respectively, although a statistical examination of the data revealed that the difference was insignificant. This indicated that vibration affects the reactivity of the organism to these drugs. In the second series, the ability of control and vibrated animals to adapt to hexanol (100 mg/kg) was tested (65 mice). The preparation was IP injected after 15 min or 4 hr of vibration, as well as on a daily basis thereafter. Table 1 shows the re-

Table 1. Duration of the anesthetic effect of hexanol on control and vibrated mice (mean duration by group in min)

Experimental action	No. of mice	Days of hexanol injection				
		1st	2nd	3rd	4th	5th
Hexanol alone	31	45	50	29	31	32
15 min of vibration prior to 1st hexanol administration	18	106	32	28	22	80
4 hr of vibration prior to 1st hexanol administration	18	110	32	27	31	48

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sults of this test. Three days after this test, the animals were given a toxic dose of strychnine (1.5 mg/kg) which was fatal for control mice in 50% of the cases. Mortality for animals which had been exposed to vibration 15 min or 4 hr prior to hexanol administration was 52% and 75%. For mice given hexanol alone, the mortality was 56%. The difference in mortality between these groups was found not to be statistically

Table 2. Swimming duration of control and experimental mice

Test no.	Experimental action	No. of mice	Swimming duration, min (± m)	Reliability		
				Rel. to test 1	Rel. to test 3	Rel. to test 4
1	Control	20	278±12,0	—	—	—
2	Vibration, no cystamine	20	272± 0,5	0,4	—	—
3	Cystamine, no vibration	28	165± 0,0	10,4	—	—
4	Vibration plus cystamine	28	115± 4,8	12,6	3,9	—
5	Cystamine plus vibration	30	103± 7,0	12,6	4,5	1,4

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ACC NR: AP6011437

reliable. To test the effects of vibration and cystamine on the working ability of the organism, mice were exercised by swimming. Cystamine (225 mg/kg) was given either 15 min before or 15 min after vibration, whereupon the animals were placed in a tub of water (24±1°C) until exhaustion occurred. Animals unable to swim for 1 hr were eliminated from this test. The results of this test are given in Table 2. These data show that vibration does not decrease working ability but that cystamine given before or after vibration does. Cystamine decreased the tolerance of the organism to exercise but statistically less so than when administered in combination with vibration. Orig. art. has: 2 tables.

[CD]

SUB CODE: 06/ SUBM DATE: 29 May 65/ ORIG REF: 006/ A&D PRESS: 4/23/

Card 4/4 UV<sup>2</sup>

LAWECO-66 FEB-2/SEC(1)/SEC(k)-2 SOTB IT, DO/GI  
ACC NR: AP6007747

SOURCE CODE: UR/0293/66/004/001/0156/0161

AUTHORS: Dolone, N. L.; Yegorov, B. B.; Antipov, V. V.

ORG: none

TITLE: The effect of factors of the space flight in the manned satellite "Voskhod"  
on Tradescantia paludosa microspores

SOURCE: Kosmicheskiye issledovaniya, v. 4, no. 1, 1966, 156-161

TOPIC TAGS: microspore, cosmonaut, artificial earth satellite, satellite data analysis, mitosis, microbiology/ Voskhod artificial earth satellite, Vostok 3 artificial earth satellite, Vostok 4 artificial earth satellite, Vostok 5 artificial earth satellite, Vostok 6 artificial earth satellite

ABSTRACT: The results of a study of the effect of the factors of the space flight of "Voskhod" on Tradescantia paludosa microspores are given. Stalks of Tradescantia paludosa with racemes were placed in special holders in the satellite. The anthers were fixed after planting 4 times: 2 hrs and 15 min, and 24, 48, and 120 hrs. The buds were also fixed 1.5 hrs before planting by cosmonaut B. B. Yegorov. Mitosis in the Tradescantia paludosa microspores lasted 7 days at 30°C (interphase 5 days, early prophase 1 day, and all remaining phases 1 day) and the entire cycle lasted 10 days at 20°C (interphase 7 days, early prophase 1.5 days, and all remaining phases another 24 hrs). It was found that the late and middle prophases were the most sensitive, and

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ACC NR: AP6007747

that the early interphase was the least sensitive. This study confirmed the earlier hypothesis of N. L. Delone, V. F. Bykovskiy, V. V. Antipov, G. P. Parfenov, V. G. Vysotskiy, and N. A. Rudneva (Kosmich. issled., 2, No. 2, 320, 1964) that reorganization of the chromosomes is caused by one set of flight factors, while disruption of the mitosis mechanism is caused by another set of factors. Orig. art. has: 7 tables.

SUB CODE: 22, 06/ SUBM DATE: 02Sep65/ ORIG REF: 002

Card 2/2111LR

L 376-3-66 FSS-2/EWT(1)/EEC(k)-2/FCC/T SCITE TT/DD/JK/GW  
ACC NR: AP6024650 SOURCE CODE: UR/0216/66/000/004/0592/0593

AUTHOR: Zhukov-Verezhnikov, N. N.; Mayskiy, I. M.; Pekhov, A. P.;  
Rybakov, N. I.; Dobrov, N. N.; Antipov, V. V.; Kozlov, V. A.;  
Saksonov, P. P.; Podoplelov, I. I.

ORG: none

TITLE: Results of study of the effect of cosmic radiation and other spaceflight factors on lysogenic bacteria and human cell cultures  
(Paper presented at the Anniversary Symposium of the Institute of Biophysics of the Czechoslovak Academy of Sciences held in Brno in May 1965)

SOURCE: AM SSSR. Izvestiya. Seriya biologicheskaya, no. 4, 1966,  
592-593

TOPIC TAGS: spaceflight effect, radiation effect, HeLa cell, lysogenic bacteria / Vostok 4 spacecraft, Vostok 6 spacecraft, Voskhod 1 spacecraft

ABSTRACT: Single-layer cultures of normal human cells (fibroblasts and amniotic cells) and human cancer cells (HeLa strain), together with cultures of lysogenic bacteria (E. coli K-12), have been consistently used as radiation indicators on Soviet spacecraft. Results of these experiments have shown that repeated exposure of a culture of HeLa cells to spaceflight factors on the Vostok-4 and Vostok-6 flights produced

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ACC NR: AP6024650

changes in experimental cells as compared with laboratory controls and with HeLa cells exposed on one spaceflight only. A longer latent period of recovery of growth capacity and other characteristics [not named] were noted in twice-flown cultures. In addition, the coefficient of proliferation for HeLa cells exposed on both Vostok-4 and Vostok-6 was one-half that for intact controls and for HeLa cells exposed to spaceflight only once. These data suggest that spaceflight factors have a cumulative biological effect on human cell cultures. However, a direct dependence of biological effect on length of spaceflight exposure has not been established in experiments with the other radiation indicator, the lysogenic bacteria E. coli K-12 ( $\lambda$ ). It is interesting to note that when the same HeLa cells used on Vostok-4 and Vostok-6 were also exposed on Voskhod-1, a well-defined drop in the proliferation coefficient was observed in comparison with intact cultures. Experimental colonies were more compact, and there were more dead cells. Other reliable differences [not enumerated] were also found between intact controls and thrice-exposed cultures. However, no reliable differences could be detected between thrice-exposed HeLa cells and a control strain used only on Vostok-6. It is suggested that the biological effect of spaceflight may be the result of the combined influence of radiation, vibration, and weightlessness. [JS]

SUB CODE: 06/ SUBM DATE: none/ ATD PRESS: 5/46

Card 2/2 vmb

ACC NR: AP6028342

SOURCE CODE: UR/0293/66/004/004/0610/0633

AUTHOR: Volynkin, Yu. M.; Antipov, V. V.; Davydov, B. I.; Dobrov, N. N.;  
Nikitin, M. D.; Pisarenko, N. F.; Saksonov, P. P.

ORG: none

TITLE: Assurance of radiation safety during the Voskhod-1 and Voskhod-2 flights

SOURCE: Kosmicheskiye issledovaniya, v. 4, no. 4, 1966, 630-633

TOPIC TAGS: space radiation, ~~radiation safety~~, solar flare, ~~ionization~~, radiation shielding, radiation dosimetry, nuclear emulsion, radiation ~~survival~~, EVA, lysogenic bacteria/Voskhod-1 Voskhod-2 ~~spacecraft~~

ABSTRACT: The Voskhod-1 and Voskhod-2 flights were characterized by extremely high orbits (apogee 495 km). It was calculated that Voskhod-2 would have a far higher radiation exposure due largely to the proton component in the area of the Brazilian anomaly, where in the course of 20 min the spaceship would acquire about 80% of the daily dose. The extravehicular surface dose of electrons during 20 min could amount to 1 rad. In order to reduce this to zero a protective layer of 100 mg/cm<sup>2</sup> is required. Leonov's spacesuit fulfilled this shielding requirement. Since exposure to radiation may reach dangerous proportions during solar flares the following radiation protection measures were taken during the Voskhod-1 and Voskhod-2 flights. A preliminary study was made of radiation conditions on the proposed orbit. Forecasts

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UDC: 614.876(202)

ACC NR: AP6028342

of the possibility of solar flares were made. The radiation dose was reduced by spacecraft shielding. Changes in the level of radiation in the upper atmosphere were checked by means of balloon sondes. Integral doses and dose rates were measured by on-board radiation meters. Individual dosimeters of the ILK, IKS, and IFKN types and nuclear emulsions were used to measure the total doses acquired by each cosmonaut. Living organisms were carried on board as biodosimeters. Radioprotective drugs were carried for emergency use by the cosmonauts. In order to determine the effect of low-energy electrons during Leonov's EVA the two cosmonauts carried identical sets of dosimeters (on the chest under the spacesuit and in external hip pockets), which were capable of working in high-vacuum conditions. However, Leonov's dose did not exceed Belyayev's. Individual and on-board dosimeters indicated that the total dose received on Voskhod-2 was  $70 \pm 5$  mrad, while that on Voskhod-1 was  $30 \pm 5$  mrad. Analysis of the spectral composition of radiation made by nuclear emulsions indicated the presence of particles with linear energy losses comparable to ions of He, B, O, and Ar. The radiation dose, taking RBE into account, did not exceed several dozen rads. Biological objects carried on Voskhod-1 and Voskhod-2 showed increases in non-disjunction of chromosomes and increases in frequency of dominant lethal mutations in Drosophila, and disruption of the mitotic mechanism in microspores of Tradescantia; these increases, however, were small. Lysogenic bacteria carried on the two Voskhod flights did not show any effect of radiation or other spaceflight factors. Experiments performed by B. B. Yegorov have indicated that various stages of mitosis in Tradescantia microspores possess varying sensitivity to the effects of spaceflight factors. These findings confirmed Yegorov's hypothesis that the chief cause of

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disruption of the mitotic mechanism is weightlessness and that chromosome reconstructions are due largely to combined factors related to spaceflight takeoff and reentry. Orig. art. has: 2 tables. [BM]

SUB CODE: 06/ SUBM DATE: 21Aug66/ ORIG REF: 006/ ATD PRESS: 5064

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CIA-RDP86-00513R000101720009-1"

ACC NR: AT6036477

SUTB DD/GD

SOURCE CODE: UR/0000/66/000/000/0030/0031

AUTHORS: Antipov, V. V.; Kozlov, V. A.; Davydov, B. I.; Dobrov, N. N.;  
Razgoverov, B. L.; Saksanov, P. P.

31

B+1

ORG: none

TITLE: New data on changes in the reactivity of the organism under the effect of several spaceflight factors [Paper presented at the Conference on Problems of Space Medicine held in Moscow from 24-27 May 1966]

SOURCE: Konferentsiya po problemam kosmicheskoy meditsiny, 1966. Problemy kosmicheskoy meditsiny. (Problems of space medicine); materialy konferentsii, Moscow, 1966, 30-31

TOPIC TAGS: space physiology, combined stress, biologic vibration effect, biologic acceleration effect, ionizing radiation biologic effect, rat, cystamine, strychnine, proton radiation biologic effect

## ABSTRACT:

Experiments were performed to test changes in the reactivity of the organism which result from spaceflight factors (vibration, acceleration, ionizing radiation) and their combinations. The functional condition of the organism was evaluated using pharmacological and physical methods.

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It was found that vibration (70 cps at 10 G, for 1 hr) did not affect the stamina of the animal to physical exercise (swimming). The administration of cystamine (225 mg/kg) either before or after vibration caused a marked decrease in the duration of the swimming by the animal. Cystamine alone decreased the stamina of the organism during exercise, but to a significantly smaller degree than in combination with vibration. Vibration had the effect of moderately increasing the sensitivity of the organism to cystamine (400 mg/kg) and strychnine (1.5 mg/kg).

Four hours after exposure to acceleration (8 G, chest-back, for 20 min), a statistically significant drop in the physical stability of the animals was observed. On the seventh day after exposure stability increased. Changes in the reactivity of centrifuged animals with respect to physical exercise corresponded to shifts in the ceruloplasmin in the blood.

Forty days after exposure to protons (energy 120 Mev, doses from 700--1770 rad), the stability of animals to physical loads was lowered. Preliminary centrifugation (8 G for 15 min four hours prior to irradiation with doses of 400 and 700 rad) increased somewhat the resistance of animals to radiation. [W. A. No. 22; ATD Report 66-116]  
SUB CODE: 06 / SUBM DATE: 00May66  
Cord 2/2

ACC NR: A16036632

SOURCE CODE: UR/0000/66/000/000/0335/0336

AUTHOR: Saksonov, P. P.; Antipov, V. V.; Dobrov, N. N.; Kozlov, V. A.; Shashkov, V. S.

ORG: none

TITLE: Problems of pharmacochemical protection of the organism against ionizing radiation on spaceflights [Paper presented at the Conference on Problems of Space Medicine held in Moscow from 24-27 May 1966]

SOURCE: Konferentsiya po problemam kosmicheskoy meditsiny, 1966. Problemy kosmicheskoy meditsiny. (Problems of space medicine); materialy konferentsii, Moscow, 1966, 335-336

TOPIC TAGS: radiation protection, pharmacology, ionizing radiation biologic effect, cosmic radiation biologic effect, life support system, radiation tolerance, space medicine

ABSTRACT:

Although some pharmacochemical substances have a demonstrated ability to increase the radioresistance of both humans and animals, they cannot be used unconditionally in spaceflight. Special features of the cosmic radiation effect which must be considered in the search for effective

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ACC NR: AT6036632

radioprotective agents include: 1) the complex spectrum of cosmic radiation and its variable dose power (Protons of varying energies and undetermined RBE comprise 80% of cosmic radiation); and 2) alteration of the organism's reactivity to chemical substances, and to the combined effect of radiation and other spaceflight factors, such as acceleration, weightlessness, and altered pressure. Unfortunately, effective radioprotectors alter the organism's reactivity in such a way that it becomes less resistant to the effects of unfavorable flight factors, especially acceleration and vibration.

Before the problem of human pharmacocochemical protection in spaceflight can be solved, a number of important studies must be conducted. First, the possibility of use of antiradiation agents during irradiation of biological objects with low, variable dose powers must be determined. Much more information about the effectiveness of such drugs under the combined influence of radiation and other spaceflight factors is necessary. Then it may be possible to eliminate the unfavorable effect of radioprotectors on the organism's tolerance to other spaceflight factors.

Radioprotective substances intended for use in spaceflight must offer good protection without altering the organism's resistance to other spaceflight factors, even after multiple administration. They must not hinder

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work capacity even briefly, and also must be available in convenient medicinal form. In addition, radioprotectors used in spaceflight must not damage the hereditary structures or disrupt the physiological functions of links in the spacecraft life-support system.

(W.A. No. 22; ATD Report 66-116)

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L 03777-67 FSS-2  
ACC NR: AP6028343

SOURCE CODE: UR/0293/66/004/004/0634/0640

ACC NR: AP6028343  
AUTHOR: Zhukov-Verczhnikov, N. N.; Mayskiy, I. N.; Delone, N. L.; Rybakov, N. I.;  
Kozlov, V. A.; Davydov, B. I.; Antipov, V. V.; Saksonov, P. P.; Rybakova, K. D.;  
Tribulev, G. P.

ORG: none

ORG: none  
TITLE: Biological investigations on the Voskhod-1 and Voskhod-2 spaceships  
DATE: no. 4, 1966, 634-640

SOURCE: Kosmicheskiye issledovaniya, v. 4, no. 4, 1968, p. 11.

**TITLE:** Biological investigation of the influence of spaceflight on bacteria, yeast, and mold in Voskhod-1 and Voskhod-2 spaceships

**ABSTRACT:** Experiments were performed on the Voskhod-1 and Voskhod-2 spaceships to test the effects of spaceflight on lysogenic cultures of *E. coli* K-12 ( $\lambda$ ). The cultures were carried in 1.5-ml ampules on board spaceships and in Leonov's spacesuit pocket during his EVA. Some of the ampules contained the radioprotective drug S-mercaptopropylamine. Controls were kept at the cosmodrome and at the home laboratory. Results showed that on the basis of viability there was no difference between samples carried on Voskhod-1 and the controls. Experiments on Voskhod-2 resulted in a slightly higher viability on the part of experimental cultures as compared to controls. Phage production of experimental cultures carried on the two flights also did

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, not exceed phage production of controls. Thus, it was not possible to demonstrate the protective properties of  $\beta$ -mercaptopropylamine. An attempt was made to determine whether spaceflight sensitized lysogenic cultures of *E. coli* K-12 (1) to consequent exposure to small doses of x-rays. Results showed that phage production in space-flown samples was almost identical to that of the controls. In addition, air-dried seeds of pine and winter wheat (PPG-186) were carried on *Voskhod-2* and in Leonov's pocket during his EVA for the purpose of determining the genetic effects of space-flight factors. Results did not reveal any substantial differences between the two spaceflight-exposed groups of seeds and the controls. It is assumed that the absence of the effects of spaceflight factors on lysogenic bacteria and seeds of higher plants in these two flights is due to the particular conditions under which these flights took place. Orig. art. has 5 tables. [BM]

SUB CODE: 22 SUBM DATE: 21Apr66/ ORIG REF: 013/ OTH REF: 002/ ATD PRESS: 5063

Cord 2/2 14

ACC NR: AT6036563

SOURCE CODE: UR/0000/66/000/000/0172/0173

AUTHOR: Zhukov-Voroshnikov, N. N.; Mayskiy, I. N.; Tribulev, G. P.; Rybakov, N. I.;  
Podoplelov, I. I.; Dobrov, N. N.; Antipov, V. V.; Kozlov, V. A.; Saksonov, P. P.;  
Parfenov, G. P.; Sharyy, N. I.

ORG: none

TITLE: Some results and trends in the study of the biological effect of cosmic  
radiation and dynamic flight factors using microbiological and cytological models  
[Paper presented at the Conference on Problems of Space Medicine held in Moscow from  
24 to 27 May 1966]SCURCE: Konferentsiya po problemam kosmicheskoy meditsiny, 1966. Problemy  
kosmicheskoy meditsiny. (Problems of space medicine); materialy konferentsii,  
Moscow, 1966, 172-173TOPIC TAGS: manned space flight, space biologic experiment, tissue culture, lysogenic  
bacteria, cosmic radiation biologic effect, combined stress/Voskhod-1ABSTRACT: Systems of lysogenic bacteria and single layer cultures of normal and  
cancer cells of man have been used on all spaceflights since the second orbital  
spaceship. This report presents the results of investigations performed on  
spaceships of the Vostok and Voskhod types. Biological experiments  
carried out on Vostok-3, -4, -5, and -6 indicate that phage production  
of lysogenic culture of E. coli K-12 increases with the duration of the  
flight. However, a direct linear relationship between the biological

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effect and the time of exposure in space was not established. The results obtained make it possible to assume that the biological effect in the above experiments depends on the combined effect of spaceflight factors, and specifically vibration, weightlessness, and radiation.

Ground experiments have indicated that the sensitivity of a lysogenic bacteria system to gamma irradiation ( $\text{CO}^{60}$ ) increases if the bacteria were previously exposed to vibration. These results not only confirm this supposition but make a more differentiated approach to evaluation of various spaceflight factors possible. However, in order to obtain a more complete picture of the genetic and radiation hazard of such flights, it is necessary to consider data obtained with more highly organized biological objects. Consequently, the results of spaceflight experiments performed with single-layer cultures of somatic human cells are of definite interest. In the series of experiments carried out on Vostok-1, -2, and -4, it was found that viability, and such indices as the coefficient of proliferation, the percentage of dead cells, and the morphological, antigenic, and cultural properties of the tissues, did not differ substantially from controls which were kept at the cosmodrome or the laboratory.

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However, when tissues were subjected to a second spaceflight (on Vostok-4, Vostok-6, and Voskhod-1), the twice-flown tissues showed a definite prolongation in the latent period of the ability to grow, as well as certain other noticeable changes. This makes it possible to surmise that spaceflight factors may have a cumulative effect on human tissue cultures. Further investigations of the biological effects of spaceflight utilizing lysogenic bacteria and tissues of various cultures are contemplated. [W.A. No. 22; ATD Report 66-116]

SUB CODE: 06, 22 / SUBM DATE: 00May66

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ACC NR: AF7011642

SOURCE CODE: UR/0000/66/000/000/0001/0006

AUTHOR: Volynkin, Yu. N.; Antipov, V. V.; Davydov, B. I. Dobrov, N. N.;  
Nikitin, M. D.; Pisarenko, N. F.; Saksorov, P. P.

ORG: none

TITLE: Radiation safety during the flights of the Voskhod and Voskhod-2  
spaceships

SOURCE: International Astronautical Congress. 17th, Madrid, 1966. Doklady.  
no. 4. 1966. Obespecheniye radiatsionnoy bezopasnosti pri poletakh korabley  
"Voskhod" i "Voskhod-2", l-6

TOPIC TAGS: ionizing radiation biologic effect, proton radiation biologic  
effect, EVA, space physiology, space biologic experiment, space flight /  
Kosmos-47 space flight, Voskhod-1 space flight

ABSTRACT:

Radiation conditions on the Voskhod-1 trajectory  
were forecast using Kosmos-47, a satellite launched  
into the proposed orbit shortly before the manned space-  
flight. A greater radiation hazard was predicted for

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the Voskhod-w spacecrew because of the higher orbit and extravehicular activity planned for this flight. Preliminary calculations set the maximum 24-hr dose at 0.1 rad, about 80% of which was expected to accumulate during 20 min spent passing through the region of the Brazilian anomaly. It was calculated that the EVA would expose Leonov to as much as 1 rad of electron radiation in a 20-min period, and that shielding of 100 mg/cm<sup>2</sup> would be required to eliminate this hazard. Leonov's spacesuit fulfilled the shielding requirement. A total dose of no more than several dozen REM was anticipated for the Voskhod spacecrew for the 24-hr period.

The possibility of radiation injury from solar flare protons was carefully considered. Disruptions of the earth's geomagnetic field after some solar flares are known to affect the "radiation screen" of the geomagnetic field. Thus, approximate total doses from large flares of the type 10 March 1959 and 12 November 1960 were calculated with different shielding thicknesses, discounting the screening effect of the Earth's magnetic field. (see Table 1)

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Table 1

Energy of protons $E$ , Mev	Shielding of air-equivalent substance, g/cm <sup>2</sup>	Dose from flare, rad	
		Nov. 12, 1960	May 10, 1959
$E > 40$	1.5	550	1120
$E > 80$	5.0	90	70
$E > 100$	7.0	50	20
$E > 200$	24.0	10	1

As can be seen from the table, cosmonauts can receive radiation doses sufficient to disrupt working capacity or endanger life during a solar flare. Consequently, an important part of the radiation safety program consists of predicting potentially hazardous solar flares.

In addition to the measures just described, the Voskhod radiation safety system included measurements of radiation levels in the upper atmosphere using sounding balloons. In addition, a radiometer on the craft measured total dose and dose rate, each cosmonaut carried

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individual dosimeters (ILK, IKS, and IPKN types, and nuclear emulsions), and there were biological dosimeters on board. Chemical radioprotectors were available for emergency situations.

In order to determine the possible effect of electron radiation during the EVA, both cosmonauts wore an identical set of dosimeters equipped to work in a vacuum, one in the chest area under the suit, and one in the outside hip pocket.

Although the period before the Voskhod-1 launch was one of minimal solar activity, on October 9, 1964, (3 days before the launch) at 8:30 A. M. a 23-fold increase in radioactivity was noted in the upper atmosphere at an altitude of 22 km. The increased radioactivity lasted 2 hr and is still unexplained.

Doses obtained by Voskhod crew members are shown in Table 2.

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ACC NR. AT/01162

Table 2. Total radiation doses obtained by crew members on Voskhod-1 and Voskhod-2 spacecraft, in mrad (tissue)

Name of spacecraft	Individual dosimeters		R-ZAM on-board dosimeter	
	average dose for flight, mrad	average dose rate, mrad/day	dose for flight, mrad	dose rate, mrad/day
Voskhod-1	30±5	29±3	27±1	26±1
Voskhod-2	70±5	65±3	65±1	60±1

The total radiation dose received by Leonov was not higher than that obtained by Belyayev due to electron radiation outside the spacecraft, as had been expected. The fact that the absorbed tissue doses received by Voskhod-1 and Voskhod-2 cosmonauts were two and four times higher, respectively, than doses received on the Vostok flights can be explained by the difference in orbits and by some increase in the intensity of primary cosmic radiation characteristic for quiet Sun periods.

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The radiation doses for Voskhod crew members did not exceed several dozen REM each, as calculated. This radiation dose is not considered injurious to human health.

Biodosimeters carried on the Voskhod craft included seeds of higher plants, microorganisms, and fruit flies. In addition, Leonov had pine and wheat seeds and lysogenic bacteria in his hip pocket during the EVA. Analysis of this biological material showed that spaceflight factors had the following effects: mitosis was disrupted in *Tradescantia paludosa* microspores, and there were more dominant lethaliities and cases of nonseparation of chromosomes in *Drosophila*. These shifts were of the same type as those observed in the Vostok-2, -3, and -6 experiments, and were also numerically insignificant. Lysogenic bacteria and plant seeds exposed in open space or kept in the spacecraft did not show the effects of spaceflight factors.

Yegorov's experiment with *Tradescantia* microspores demonstrated that the various mitotic phases of this organism have different sensitivities to spaceflight

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factors. Furthermore, this experiment suggested that weightlessness may be the cause of disrupted mitosis in *Tradescantia microspores*, and that chromosome rearrangements are chiefly caused by factors associated with launch and descent.

Results of biological experiments conducted on the Voskhod spacecraft are in agreement with data from physical dosimeters. Periodic postflight examinations of all Voskhod crew members have also demonstrated the absence of a harmful radiation effect. Orig. art. has: 2 tables.

ATD PRESS: 5098-F

SUB CODE: 06 / SUBM DATE: none / ORIG REF: 006

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ANTIPOV, V. Ye.

32357

TALINOV, Sh. T. i ANTIPOV, V. Ye. Ob Otkrytii Malykh Kolichesstv Zn, Ni, Co, Cu, Cd,  
i Ni v Prisutstvii Bol'shikh Kolichesstv Khroma Doklady Akad. Nauk SSSR 1/3 CCP, 1949, N<sup>o</sup> 2,  
16-17 Ryezgumye Na Uzbyek. Yez.--Bibliogr s. 15

SO: Letopis' Zhurnel'nykh Statey, Vol. 44

ANTIPOV, V. V.

Line  
3

Chemical Abst.  
Vol. 48 No. 4  
Feb. 25, 1954  
Inorganic Chemistry

New method of synthesis of chromium fluoride. Sh. T.  
Talipov and V. B. Antipov (Inst. Chem. Acad. Sci. Uzbek.  
S.S.R.). Doklady Akad. Nauk UzSSR, 1949, No. 12,

27-0 (in Russian).—Pure CrO<sub>3</sub> (40 g.), 240 g. 10% HF, 17 g.  
glucose, and 30 ml. EtOH, added in that order with agitation,  
until a spot test on filter paper shows green edges, gave  
 $\text{CrF}_3 \cdot 3\text{H}_2\text{O}$  in the soln. The latter was concd. on a steam  
bath to a paste, stirred with 50 ml. warm H<sub>2</sub>O, kept over-  
night, filtered through a Pt Büchner funnel, washed with  
EtOH and cold H<sub>2</sub>O, yielding 85-90% of product after 2  
recrystn. Purity of 99.60% was established by analysis.

G. M. Kowlapoff

MF  
1-21-5

TALIPOV, Sh.T.; ANTIPOV, V.Ye.

Investigation of solubility in the system chromium fluoride-sodium fluoride-water at 30°. Trudy Inst. Khim., Akad. Nauk Uzbek. S.S.R. 3, 206-13 '52.  
(CA 47 no.22:11930 '53) (MLRA 6:3)

1. Uzbek Acad. Sci., Tashkent.

TALIPOV, Sh.T.; ANTIPOV, V.Ye.

Separation of large quantities of chromium from small quantities of zinc,  
copper, nickel, cobalt, and cadmium. Trudy Inst. Khim., Akad. Nauk Uzbek.  
S.S.R., Inst. Khim. 3, 214-22 '52. (MLRA 6:3)  
(CA 47 no.22:12106 '53)

TALIPOV, Sh.T.; ANTIPOV, V.Ye,

Investigation of solubility in the system aluminum fluoride-lithium fluoride-water at 25°. Trudy Inst. Khim., Akad. Nauk Uzbek. S.S.R. Inst. Khim. 3, 223-9 '52.  
(MLRA 6:3)  
(CA 47 no.22:11929 '53)

ACC NR: AM6029197

Monograph

UR/

Asee, Boris Abramovich; Zhukova, Nina Mikhaylovna; Antipov, Yevgeniy Fedorovich

Parts and units of aviation instruments and their design (Detali i uzly aviatzionnykh priborov i ikh raschet) 2d ed., rev. and enl. Moscow, Izd-vo "Mashinostroyeniye", 1966. 415 p. illus., bibliog., tables. Textbook for students at aviation engineering technical schools. 15,000 copies printed.

TOPIC TAGS: auxiliary aircraft equipment, aircraft engine instrument, aircraft flight instrument, shock absorber, gyroscope

PURPOSE AND COVERAGE: The book is intended as a textbook for students of aircraft building technical schools and the aircraft instrumentation industry staff. Detailed information is presented on theory and design in the field of aircraft instrumentation and accessories such as sensor elements, mechanical transmitters, and amplifying mechanisms, components of gyroscopic instrumentation, current-carrying and computing equipment, shock absorbers, damping mechanisms, and bearings. Drawings and illustrations appear throughout the text, and formulas are given for solving various problems in the field. A brief review is given in the introduction of known contributors to the development of the Soviet aircraft instrumentation industry. Review and suggestions regarding the manuscript of this book were provided by Doctor of Technical Sciences D. A. Braslavskiy and Engineer A. A. Nikolayeva. There are 66 references, 65 of which are Soviet.

Card 1/2

UDC: 629.13.05(075)

ACC NR: AM6029197

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- Ch. I. General information on aircraft instrumentation and its operational requirements -- 5  
Ch. II. Sensor elements -- 14  
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SUB CODE: 01/

SUM DATE: 26Feb66/

ORIG REF: 065/

OTH REF: 001/

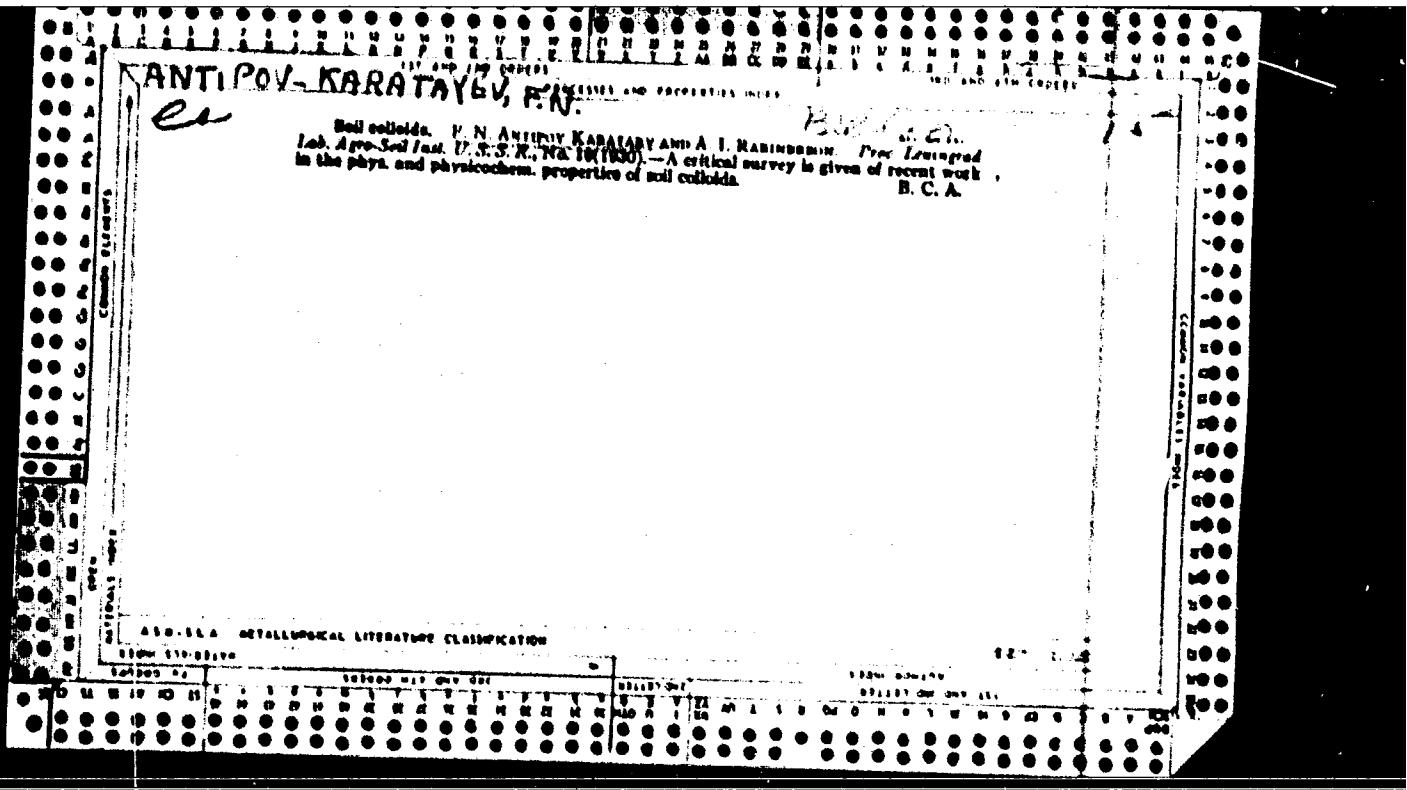
Cord 2/2

BORMOTOV, P.N., inzh.; GRISHIN, S.S.; ANTIPOV, Yu.; VITRIK, E.V., inzh.;  
KOSAREV, P.S.; NEKHOROSHEV, A.I.; RYADTSEV, G.I.; KOTOV, S.F.; SHARAGIN,  
M.A., gornospasatel' (Komi ASSR, g. Uchta)

On P.M. Solovev's article "Improve the design of the SP-55M self-rescuers." Bezop. truda v prom. 6 no.7:9-11 Jl '62. (MIR 15:7)

1. Tekhnicheskoye upravleniye Kombinata ugol'nykh predpriyatiy Kuznetskogo kamennougol'nogo basseyna (for Bormotov). 2. Master shakhty im. Lenina Makeyevskogo tresta ugol'noy promyshlennosti Donbassa (for Grishin). 3. Komandir vvedoda voyenizirovannoy gornospasatel'noy chasti, pos. Zarubino, Novgorodskoy oblasti (for Antipov). 4. Shakhta No.24, Lubanskaya oblast' (for Vitrik). 5. Zaveduyushchiy gornymi rabotami Nikitovskogo dolomitnogo kombinata (for Kosarev). 6. Komandir otdeleniya No.8 VGSO, g. Shakhty, Rostovskaya obl. (for Nekhoroshev). 7. Komandir gornospasatel'nogo otdeleniya, g. Shakhterok, Donetskaya obl. (for Ryadtsev). 8. Zamestitel' glavnogo inzh. shakhty No.29 "Kapital'naya" Chelyabinskogo kombinata ugol'nykh predpriyatiy Ministerstva ugol'noy promyshlennosti SSSR (for Kotov).

(Respirators) (Solovev, P.M.)



CA

ANTIPOV-KHAKATAEV, I.N.

The soils of the Crimean state forest sanctuary reservation and of the adjoining neighborhood. I. N. Antipov, Kharsty, and L. I. Frenzov. Transl. from the Russian. (U.S. S. R.) No. 7, 280 pp. (1933). A series of analyses—total, aq., salts, and base exchange—are presented on the mountain forest soils of Crimea, consisting of the brown forest soils and some "rendzina" at various stages of podzolization as evidenced by the replaceable H. The analyses of a no. of other soils—chernozem-like and terra rossa—are also given. I. S. Joffe

Vest. Dokuchayev Prochnaya  
Inst.

ASSISTANT METALLURGICAL LITERATURE CLASSIFICATION

**Special and methodological investigations in the laboratory of analytical and**  
**soil chemistry. I.N. ANTIPOV-KARABALY. Lenin Agr. Acad., Inst. Fertilizers**  
**Agro-Chem. Sci., Leningrad Branch (U.S.S.R.), Bull. 18, 145-50 (1952).—A check-up**  
**on the Schellackberger method for deterg. org. matter showed that a lower concn. of H<sub>2</sub>SO<sub>4</sub>**  
**(1:1 of H<sub>2</sub>O and H<sub>2</sub>SO<sub>4</sub>, or even less H<sub>2</sub>SO<sub>4</sub>) was more efficient than a 0.4 N chromic acid**  
**sola. A modification was also made of the fluorimetry (Z. engen. Chem. 81, 1-4 (1913))**  
**method for deterg. Fe<sup>++</sup>. For the formation of the complex salt, which takes out the**  
**Fe, NiI<sub>2</sub>F and K<sub>2</sub>C<sub>2</sub>O<sub>4</sub> in the ratio of 6:1 by wt. were used. To avoid the difficulty**  
**in recognising the I color methylene blue was used, which is colorless in a reduced condition.**  
**The titration with a standard I sola. was carried out as follows: to the test**  
**sola. add 12 cc. of 1:8 HCl and a few drops of methylene blue. Then add 5 g. of KH(C<sub>2</sub>O<sub>4</sub>)<sub>2</sub>,**  
**where the sola. is said, with CO<sub>2</sub> add the oxalate and fluoride. As the oxalate and**  
**fluoride dissociate, the blue color disappears and a yellowish color appears. Add the I sola.**  
**until an intense blue color appears, which, if it remains for 0.8 min., shows that**  
**the titration has been completed.**

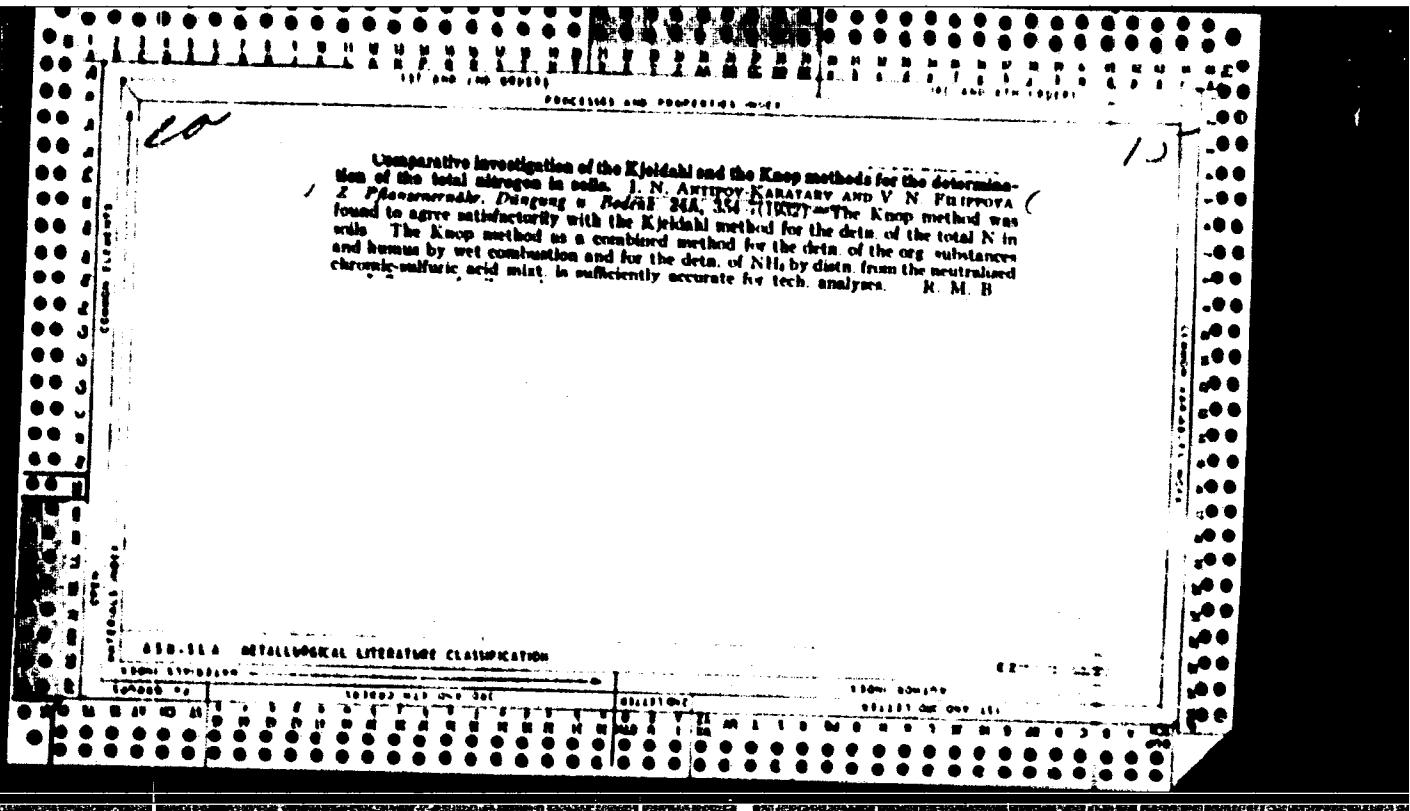
Bijl. Inst.

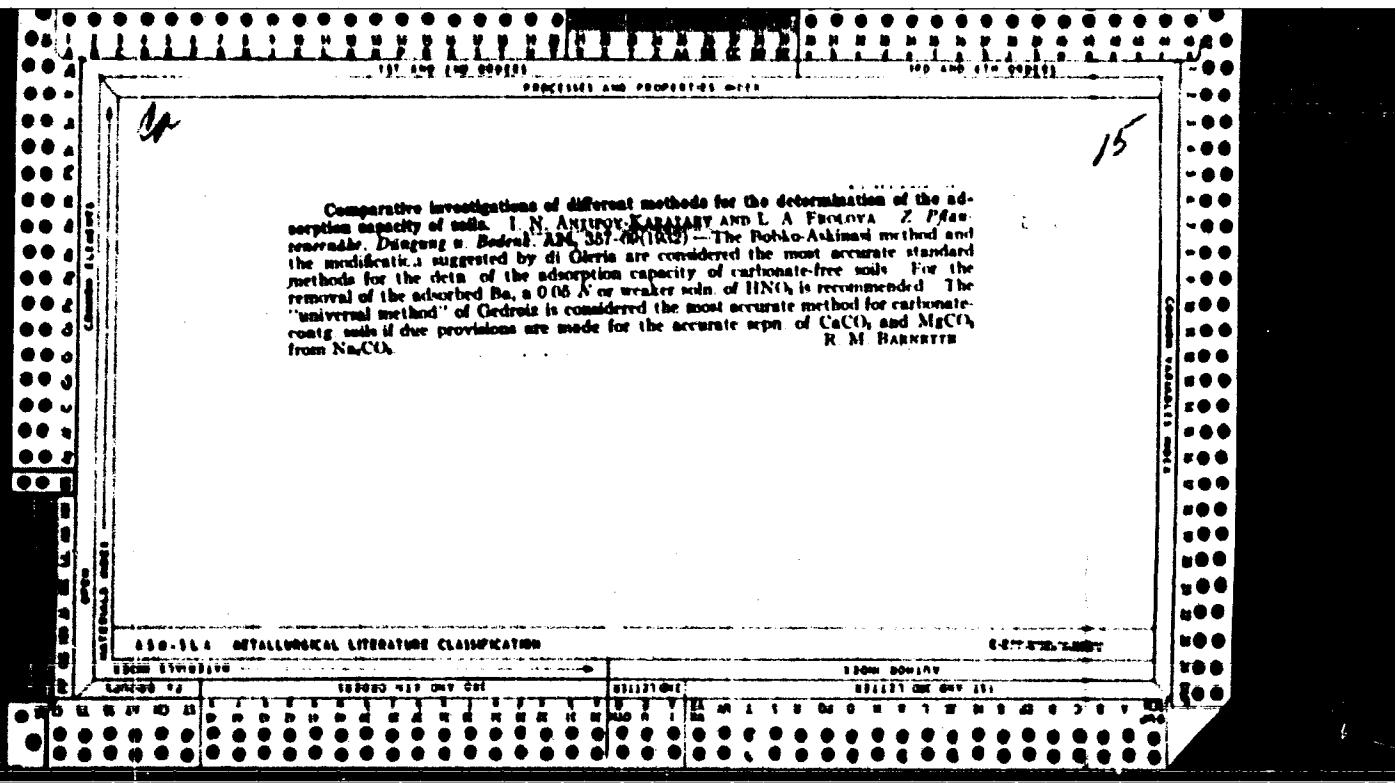
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## AMERICA METALLURGICAL LITERATURE CLASSIFICATION

**APPROVED FOR RELEASE: 06/05/2000**

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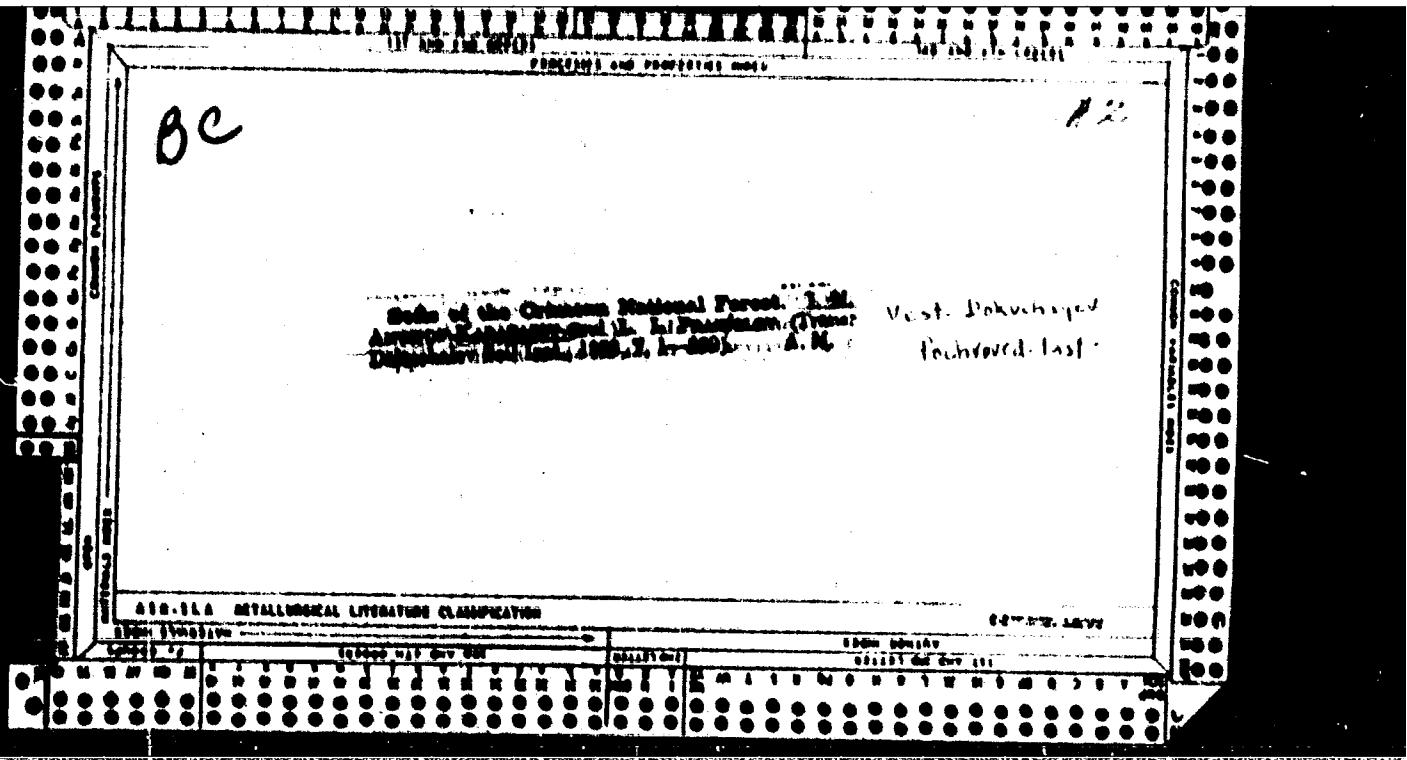
*EW*

15

REVERSAL OF SOIL CHARGE

Reversal of charge on soils by acids and anion fractions  
 I. N. Antropov-Karpenko and A. I. Rubimova. *Zem.*  
*zad. fizich. i org. Sov. Soi., Leningrad* 1952, 105-93  
 (1953) (in German).—The nature of the charge on the  
 entire soil mass was determined by electrometric methods. The  
 reversal of charge on representative Russian soil types by  
 common inorg. and org. acids was investigated. The  
 most pronounced effects were observed with the 10%  
 concn. solution of a red earth (I), peatland soils were less af-  
 fected. The charge on solonetz and chernozem soils (II)  
 was not reversed by acids. Most of the soils were af-  
 fected by HCl and  $HNO_3$ , but only I by other acids. No  
 connection with  $CaCO_3$ ;  $RuO_4$  in the fraction < 1  $\mu$  was noted,  
 but there was marked correlation of charge reversal with  
 the content of "mobile"  $H_2O_2$ , i. e., that extr. by acetate  
 or KOH. Charge reversal is attributed to a cation effect;  
 $RuO_4$  is partially dissolved in the soil, then adsorbed upon  
 particles, whence the latter become pos. The anion-  
 fixing property does not clearly parallel the ability to  
 reverse charge; for sulfate and phosphate, it was strongly  
 developed in soils not exhibiting the latter phenomenon.  
 Nevertheless, both depend upon the content of active  
 $RuO_4$ , the former also upon the nty. of salt-like combi-  
 nations formed, buffering the ntn. at an optimum not  
 below pH 2.0. Max. fixation of phosphate and sulfate was  
 observed with I, min. with II. Fixation of nitrate was  
 not observed;  $Cl^-$  was fixed by some acids irrespective of  
 ability to reverse the charge, and with one gradient and II  
 was neg. for the lower concns. of HCl. The max. was  
 with I, likewise the only acid showing adsorption of carbon-  
 ate from acid.  $CaCO_3$  to  $H_2O_2$ . There are numerous refer-  
 ences.

ASPECTS OF AGROBIOLOGICAL LITERATURE (1)

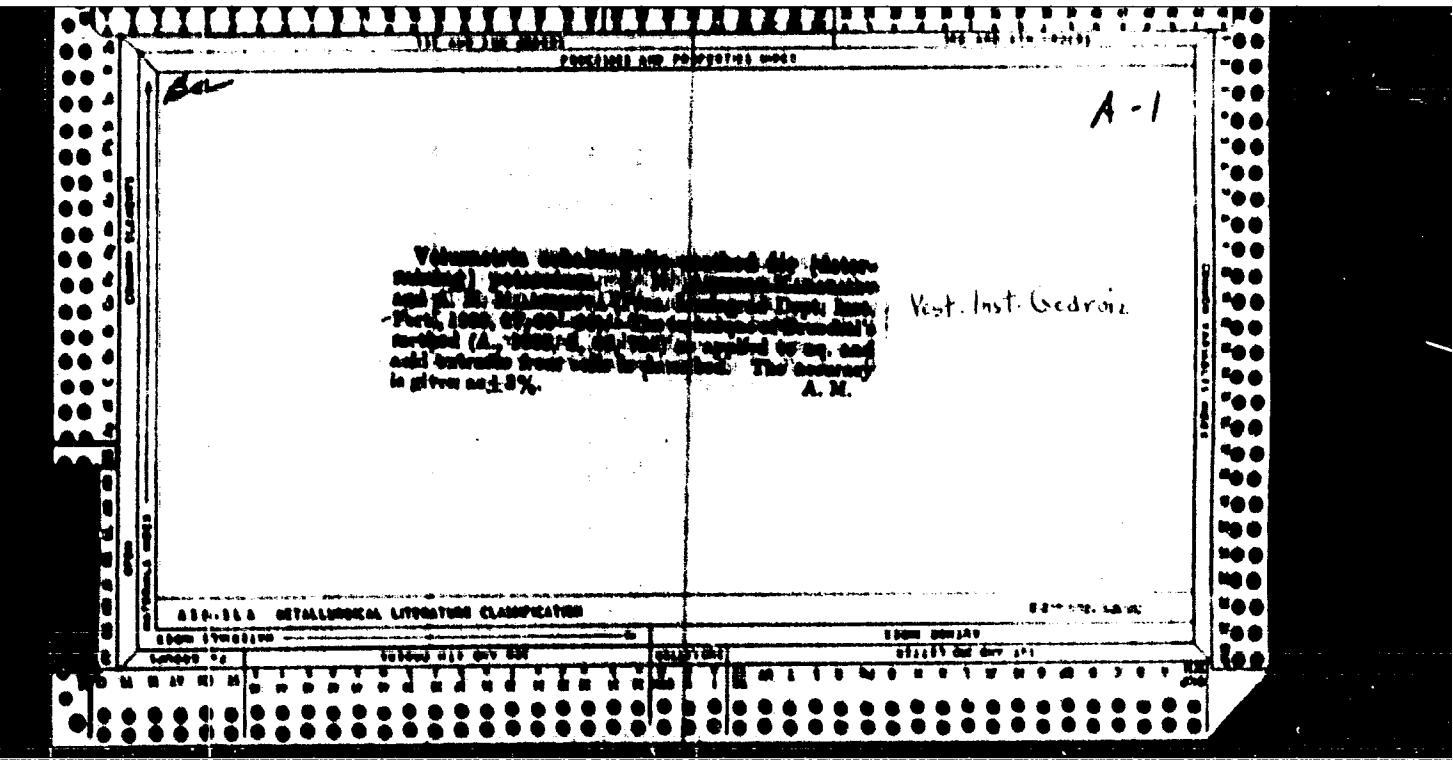


Application of electrofiltration, electrodialysis and electrolysis to soil analysis. I. I. N. Antipov-Karataev and K. N. Krasil'ev. *Trans. Doklady SSSR* 56, No. 8, 8-12 (1953).—Adsorbed Ca and Mg are removed completely by electrofiltration, Mg being removed to a greater extent than by the acetate method. The amounts of K and Na were 2-3 times those of the acetate method. This is due to the destruction of part of the soil mass. Electrodialysis gives results similar to those obtained by electrofiltration, whereas electrolysis gives results for Ca + Mg which are generally < those obtained by the acetate method.

## **Ala-11a METALLURGICAL LITERATURE CLASSIFICATION**

**APPROVED FOR RELEASE: 06/05/2000**

CIA-RDP86-00513R000101720009-1"



GC

A-1

Colorimetric picrate method for (determining)  
[REDACTED] Potassium. I. N. AGRIPIN-KARATAEV and  
A. M. MIANTRIOVA (Proc. Leningrad Dept. Inst. Fort.,  
1953, 17, 63-66). The picrate method suggested by  
Oakey (A., 1951, 427) was tested under varying con-  
ditions. Pyrolysis can be carried out at 12-15°; stir-  
ring for 10-15 min. after addition of reagent gives  
larger crystals which are best filtered through a plug  
of glass wool in an ordinary funnel. Ca<sup>+</sup>, Mg<sup>+</sup>, Al<sup>3+</sup>  
Fe<sup>3+</sup>, P, and Si do not interfere, but if Na is > twice K,  
the results are poor.

Vest. Inst

A. M.

## AIA-11-A METALLURICAL LITERATURE CLASSIFICATION

Author's name	103000-110 000 000	Editor's name	103000-000 100
103000-110 000 000	103000-000 100	103000-000 100	103000-000 100

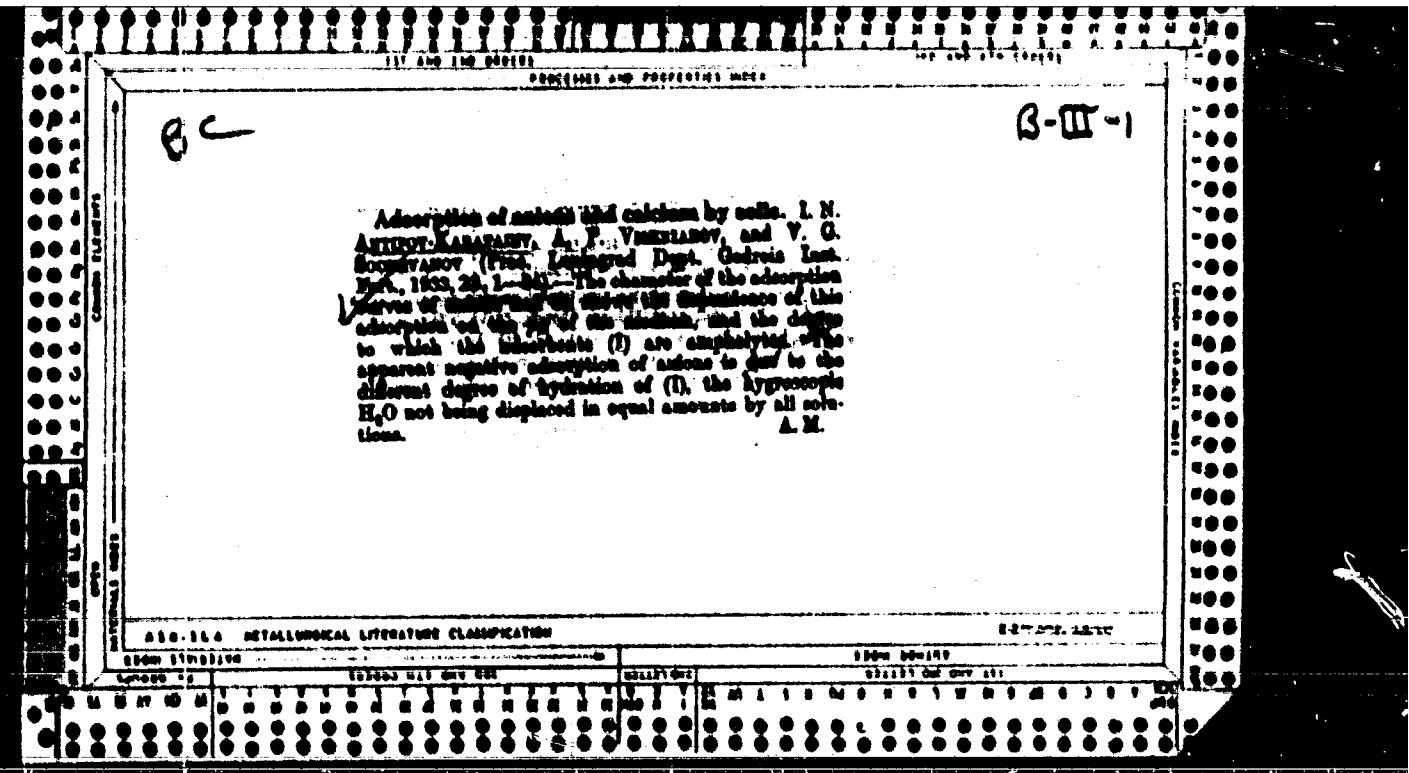
BC

B-III-1

Reversal of soil charge by acids and adsorption  
of anions. I. N. ANTIPOV-KARATAIEV and A. I.  
RADINERMAN (Proc. Landfill Sci. Dept. Inst. Petr., 1953,  
17, 101-102).—The case with which the electrical  
charge was reversed by acids ( $\text{HCl}$ ,  $\text{HNO}_3$ ,  $\text{H}_2\text{SO}_4$ ,  
 $\text{H}_3\text{PO}_4$ ,  $\text{AcOH}$ ,  $\text{H}_2\text{O}_2$ ) was, in descending order:  
Chlorite intercalate, palygorskite (Beek), red  
clay (Orlinsk). Neither smectite nor vermiculite had their  
charges reversed. Methylbenzene sulfonate reversed the charges  
in all cases. There was no strong correlation between  
 $\text{SiO}_4/\text{Al}_2\text{O}_3$  ratio of the clay ( $< 1 \mu$ ) and the ease of  
charge reversal, but there is between the latter and the  
content of mobile  $\text{Na}_2\text{O}$  (Thomas's oxalate method and  
5% KOH method of Ondroff). The reversal of charge  
depends on the adsorption of anions, and the amount  
adsorbed in an acid medium depends on the amount  
of mobile  $\text{Na}_2\text{O}$  and the solubility of the salt-like  
compounds formed with the anions. A. N.

A.D.I.S. METALLURGICAL LITERATURE CLASSIFICATION

REF ID: 4486



BC

B-3-1

RECENT RESEARCH ON SOIL CHEMISTRY IN THE U.S.S.R.  
I. N. Antipov-Karatayev (Pedology, 1934, 29, 167--170). Prelim.  
A summary of published and unpublished work.  
A. M.

ALB 1000 METALLURGICAL LITERATURE CLASSIFICATION

8-3-20-1000

*RA*

15

Vest. Dokuchayev  
Pochvoved. Inst.

Adsorption of potassium by soils. I. N. Antshov-Karapetyan and T. V. Antshova-Karapetyan. Dokuchayev Soil Inst., Studies in the Genetics and Geography of Soils 1948, 171-88 (in English). The adsorption of electrolytes by soils was studied (1) by detg. the amt. of solvent adsorbed and using this value to convert the analytical data on the initial, min., and (2) by direct data of the amt. of cations adsorbed. The following soils were studied: Verkhnechernozem, humic acid, with Ca, Chernozem soil with Na and a peatland (Viatka basin). The common method of detg. adsorption by taking the difference between the initial and equal concns. of electrolyte leads to an apparent anomalous adsorption in KCl solutions, as low as 0.1 N. This is due mainly to a partial adsorption of the solvent. An attempt to det. the amt. of solvent adsorbed by using a sugar min. and detg. the concn. of sugar initially and at equil. failed owing to "combination" of the sugar and soil in the presence of electrolytes. The direct method of measuring adsorption by analytical detn. of the amt. of cation adsorbed gave results that agree with the Pseudlich adsorption isotherm at all KCl concns. up to 0.8 N. The law of mass action, even when using activities instead of concns., can be applied only at concentrations below 0.1 N. Thirty-nine references. (Colin W. Whittaker)

An investigation of humate-forming processes by means of electrochemical methods. I. N. Antipov-Karatayev and I. A. Kholmash. Doklady SSSR 1933, 117-811 (in Russian).—The objects of this investigation were: (1) a quant. estn. of the equivalence of replacement of H ions by other cations when humates are formed and the discovery of whether this process is an exchange adsorption or a chem. reaction leading to total or partial replacement of carbonylic H in a soln. of humic acid; (2) the establishment of a quant. relation between the degree of dispersion of the humates and the nature of the combined cations; (3) the investigation of the chem. nature of humic acid, its humidity and chem. equiv., and the degree of dispersion of humic acid and of humates. Two humic acids were obtained, sol no. 1 from *Hypnum*-edge peat near Minak and sol no. 2, from young wedge peat near Leningrad, both by extn. with neutral salts after previous washing with 1% HCl to a neg. reaction for iron. The preps. were purified by simple dialysis and then by electrodialysis until a const. const. of the soln. was obtained. Preliminary cond. measurements having shown that the reaction between a soln. of humic acid and a base reaches equil. in 8-10 days, the following method was adopted. Two cc. of humic acid soln. was put into each of a series of volumetric flasks and gradually increasing volv. of a soln. of the base were added. Then, H<sub>2</sub>O was added to make the vol. up to 100 cc. Cond. /pH and dispersion data. were made after the solns.

had stood for 14 days. These cond. measurements were thus a form of titration. The bases studied were LiOH, NaOH, KOH, Ca(OH)<sub>2</sub>, Ba(OH)<sub>2</sub>, LiCO<sub>3</sub> and MgCO<sub>3</sub>; all but the last named were 0.01 N and it was 0.002 N. All titration curves show the "jump" /pH at about the same point with either alkali or alk. earth hydroxides. When titrating with alkalies a molecularly dispersed system of the humate is obtained at the moment of neutralization, the percentage of the mol. fraction (particle: less than 25μ in diam.) is nearly 100. A mol. exchange reaction, consisting of replacement by metals of all the carbonylic H of humic acid, takes place in alk. humate formation. Titration with the alk. earth hydroxides leads, on the other hand, to evaporation of more than 10% of the total humic acid at the neutral point yet the equivalence observed indicates the formation of humic acid units of a definite degree of displacement. The shape of the solid portion of the titration curves depends on the degree of hydration of the humates formed which in turn depends on the degree of hydration of the cations. In the alk. some equivalence of replacement is shown by the alkali cations but Ba and Ca hydroxides show some divergence. Analogous results were obtained for the titration of sol no. 1 with Li and Mg carbonates except that the "jump" occurs at lower pH values owing to the greater buffer action of

the carbonates. In view of the known quadrivalence of humic acid it is suggested that the existence of only one "jump" on the titration curve may be explained (1) on the assumption that the "jump" corresponds to the total replacement of the first H of humic acid, the dissociation constants of the other hydrogens being so small that the resip. curve breaks are located at  $\text{pH}$  values too high to be detected by resip., or (2) the 4 dissociation constants are nearly equal as in some other org. acids. Calcium showed the activity of the humic acid H ions to be  $1.00 \times 10^{-4}$  for sol no. 1 and  $1.00 \times 10^{-3}$  for sol no. 2. The equiv. wt. of the humic acid, taken as equal to 4 times the amt. of dry matter present per l. over the g. equiv. of H present, was found to be 320 and 320, resp., which agrees well with values in the literature. This indicates that the 4 H ions have equal dissociation constants. Cond. measurements were made at 20° on the same soils to which varying amounts of humic acid had been added and allowed to stand 14 days; the Krichbaum app. as modified by Prunda was used. Values for H-ion activity in good agreement with those obtained from the potentiometric measurements were obtained. The degree of dissociation of humic acid calcd. from the H-ion activity and the const. of titrated H indicates that it is a weak acid. The conductometric data show that, in the acid state, alk. earth ions are completely replaced by H. Na and Ca humates showed little or no hydrolysis below the neutral point but the hydrolysis of alkali humates is quite large even in the acid state. The equiv. wt. of humic acid calcd. from the position of the bend in the cond. curve varies from 818 to 820. The same values were used for dispersion measurements (the ultradilute system of Monnev was used). Peptization by alkali hydroxides and coagulation by alk. earth hydroxides terminate at the neutral point. The soils coagulate most intensely in the early stages of titration, i. e., when the acid humates are being formed. Peptization curves, however, show a max. near the neutralization point.

C. W. W.

CA

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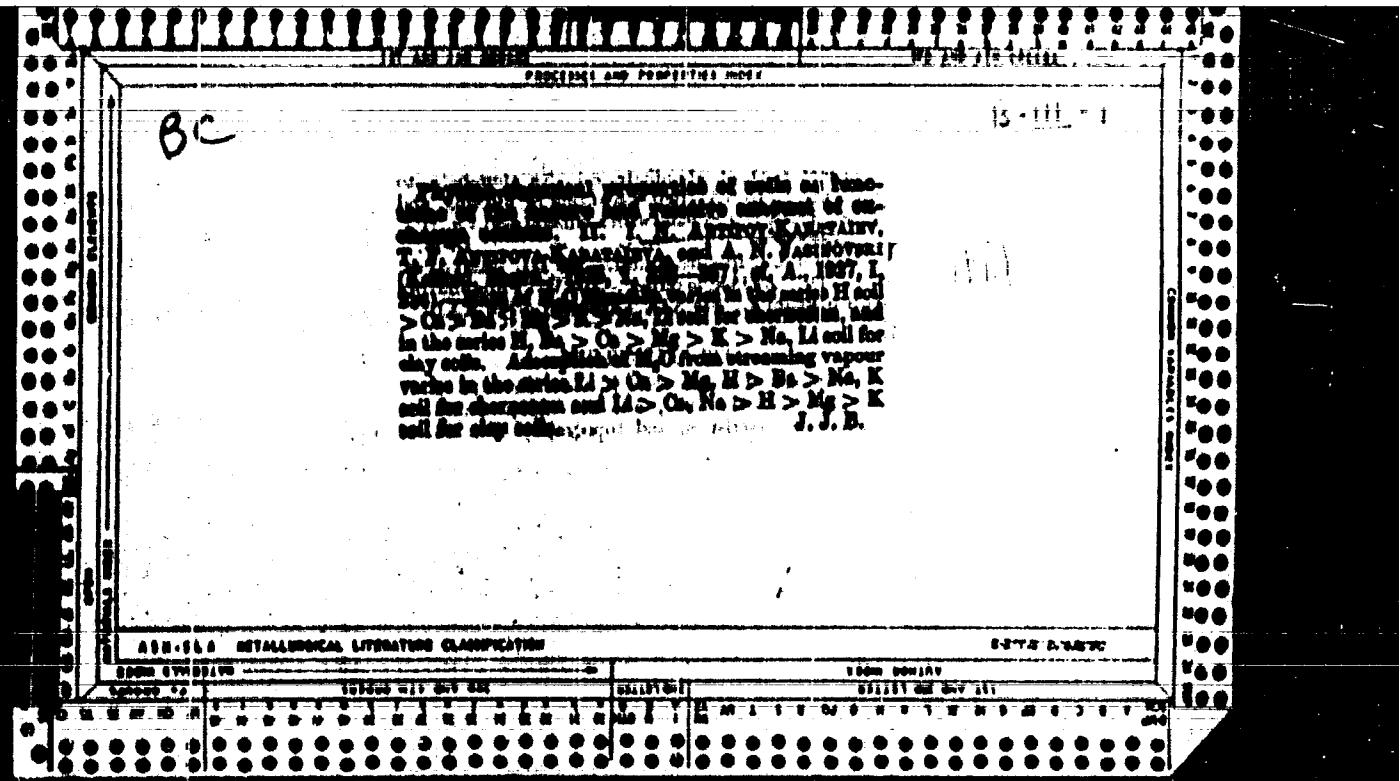
Physicochemical properties of soils as a function of  
composition and the relative exchange of cations. I

I. N. Antipov-Korolev, T. P. Antipova-Koroleva and  
A. N. Vashchenko. *Otdel. J. (U.S.S.R.)* 17:257-262  
(1938). Data are given for heats of melting and hydration, sp. gravities, densities, elec. conductivities, degree of  
saturation of colloidal complexes, cataphoretic velocity,  
method of aggregation, relative dispersion, swelling and rate of  
drainage of various U.S.S.R. soils said. by authors of  
various hums. II. Filtering capacities of soils saturated  
by various ions. *Ibid.* 33:1-87. Rates of migration and  
desorption of various ions and of water by soils said. by  
various hums are stated. F. H. Barthmann

Koll-Zher.

## APPENDIX A. METALLOGICAL LITERATURE CLASSIFICATION

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100
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*GC**d-1*

Techniques of measuring adsorption of potassium by cells. I. N. ANTIPOV-KARATAJeva (Kolloid. Zurn., 1936, 1, 481-484).—Calculation of H<sub>2</sub>O adsorbed by cells by measuring the negative adsorption of sugar from its eq. solutions is impossible in presence of electrolytes as sugar is adsorbed relatively more strongly than H<sub>2</sub>O. The adsorption of KCl can be measured only by analyzing the cell before and after the adsorption. Data are recorded and discussed.

J. J. R.

AIAA METALLURGICAL LITERATURE CLASSIFICATION

BC

8-III-1

Soil formation on the Caspian Black Sea Coast. V. V. Tikhonov, L. M. Artyuk-KARABANOV, and V. N. Slobodchikova. Acad. Agric. Nauk. Gid. Inst. Pech. No. 15. Tr. Internat. Res. Soil Inst., 1959, 49, 4). The mineralogical characteristics of these yellow podzolic soils are due to the presence of considerable amounts of secondary Fe and Al silicates and of kaolinite and smectite minerals. The silicates are probably formed in the corresponding horizons and were leached from the soil surface.

L.G.P.

## AER-11A METALLURGICAL LITERATURE CLASSIFICATION

Ergodic Division

Soviet Union

U.S.S.R.

U.S.A.

U.K.

Germany

France

Japan

Canada

Australia

New Zealand

South Africa

Brazil

Argentina

Chile

Mexico

Peru

Colombia

Venezuela

Argentina

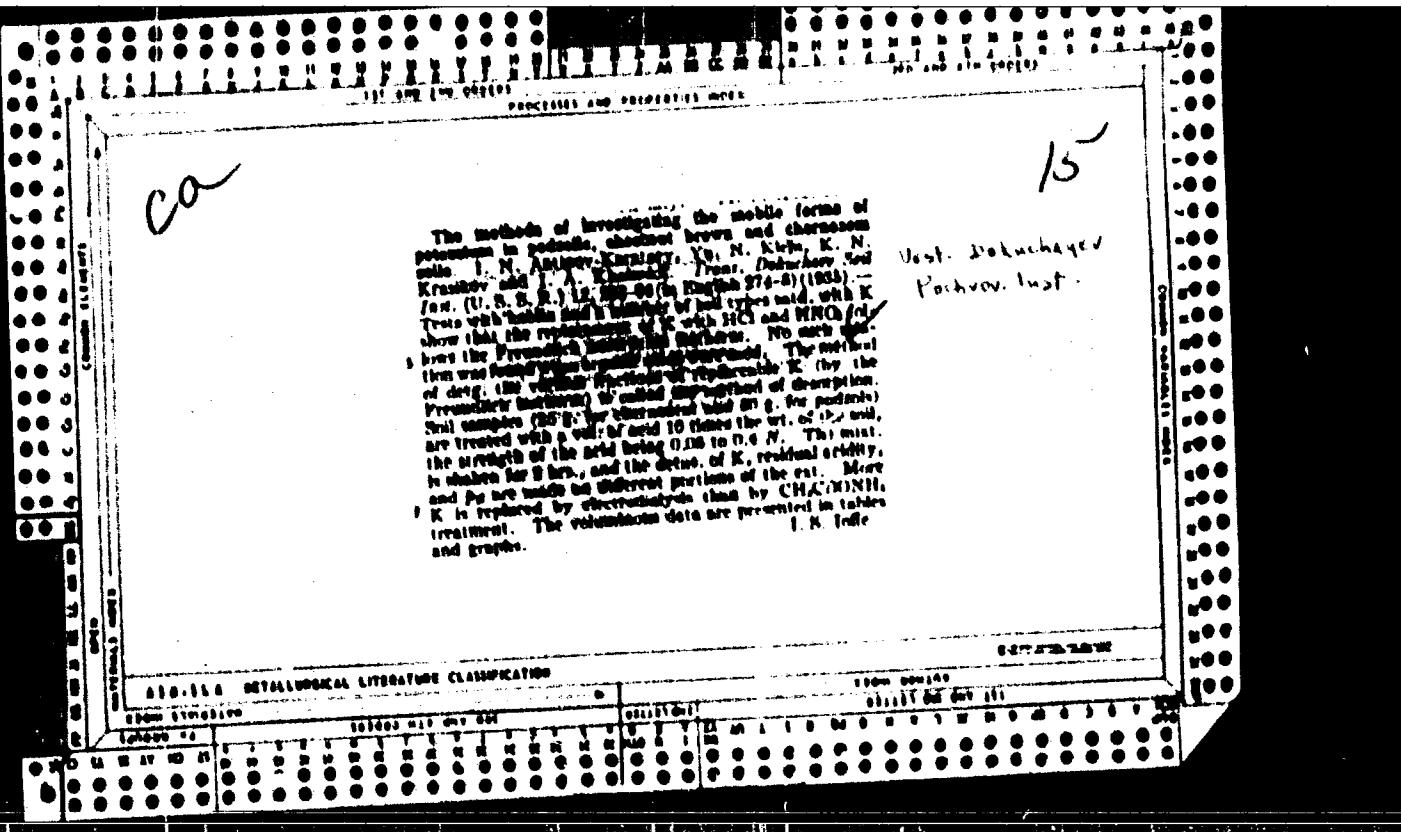
Uruguay

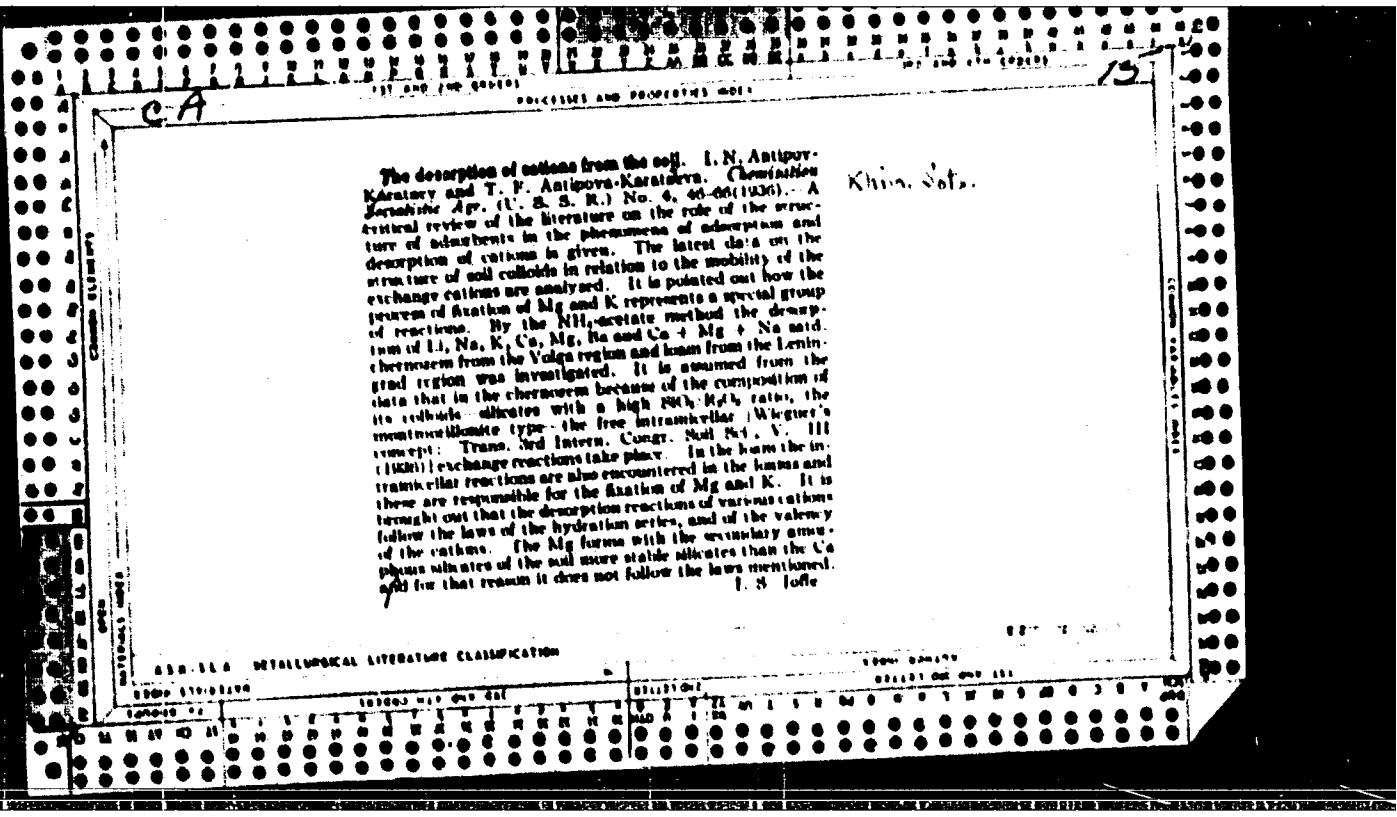
Bolivia

Paraguay

Brazil

Argentina





CA

15

Irrigation of Lower Volga soils, India. Asupur.Kare,  
Sarkar, N. Filippova and Yu. P. Stepanov. *Trans. Inst.  
Chem. Acad. SSSR*, 5, 5, 27, 6, 8-10(1968).—Irrigation  
of leached soils (soil (1:7:4 or 2 m.) of Chernozem,  
chernozem and dolomitic soils) showed that up to 15/100  
cu. m. of H<sub>2</sub>O are required to free the surface 1 m. from salin-  
salts. Mg and Na are removed by gypsum, but large  
amounts of H<sub>2</sub>O are required for complete interaction.  
Other Ca salts should be tried.

H.C.A.

## APPENDIX METALLURGICAL LITERATURE CLASSIFICATION

STANDARD CLASSIFICATION

CLASSIFICATION

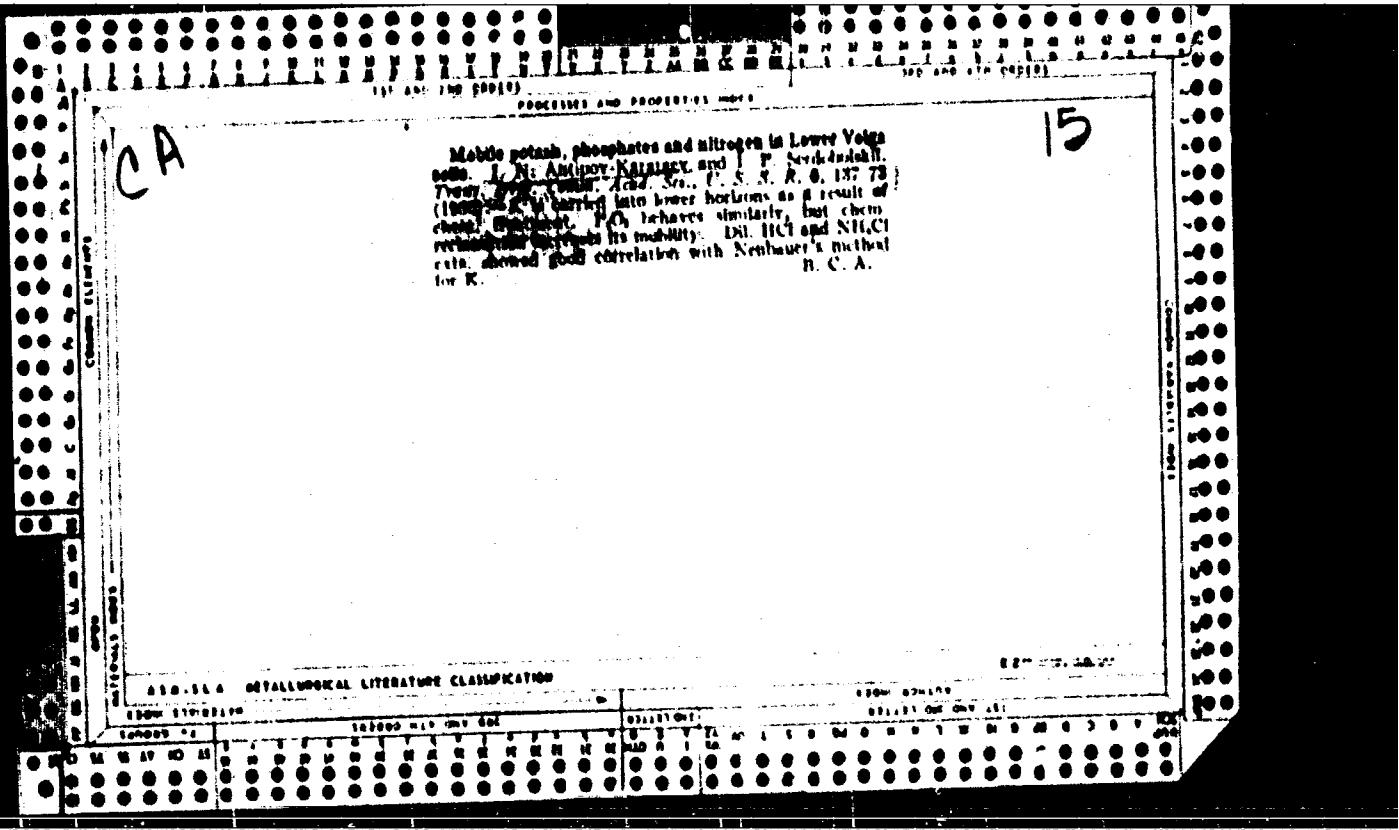
CA

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Changes in solubility of Lower Veld soils by chemical reclamation and irrigation. I. N. Antipov-Karataev and V. N. Filippova. *Vestn. Tsvetnoy Metalloobrabotki*, No. 1, 1975, p. 67-130 (1976).—Changes occurring in adsorbed humus, microaggregate analysis (cf. Vangeli, G. A. 30, 6487%), filtration analysis and data of curve of H<sub>2</sub>O infiltration proved most suitable for drug; the effect of amelioration. The action of CaO is slow; that of H<sub>2</sub>SO<sub>4</sub> is rapid but may be harmful in absence of carbonate. CaO + S proved efficient in lab. caps. R. C. A.

The equivalent weight of clay acid of soil. Shigeru Onogi and Yumio Matsumura. *J. Soil Anal. Methods*, Jan. 1972, 1-12(1972).—The equiv. wt. of clay acid from soils of 4 kinds, dried, by potentiometric and conductimetric titrations, ranged between 1.63 and 2.01. V. Kaneko

## AIA-314 METALLURGICAL LITERATURE CLASSIFICATION



*ca**15*

The mountain forest and mountain meadow soils of the Teberda Gorge, North Caucasus. J. N. Arhipov-Kostany, T. P. Antipova-Karataeva and L. I. Kuznetsova. *Voprosy Zemel'noj Sredi* /ed. (U. N. B. R.) 13, 207-04 (in English 300-7) (1936).—Data are given on the humus content, N, P, K, and exchangeable bases of the following mountain soils: podzolic brown, forest brown, meadow and mountain meadow. It is noted that in the podzolic mountain soils there is an accumulation of total humus and very little removal of soil humus. The degree of oxidation (the Springer method) is lower than in soils of the horizontal zone. J. N. Joffe.

## A8-114 METALLURGICAL LITERATURE CLASSIFICATION

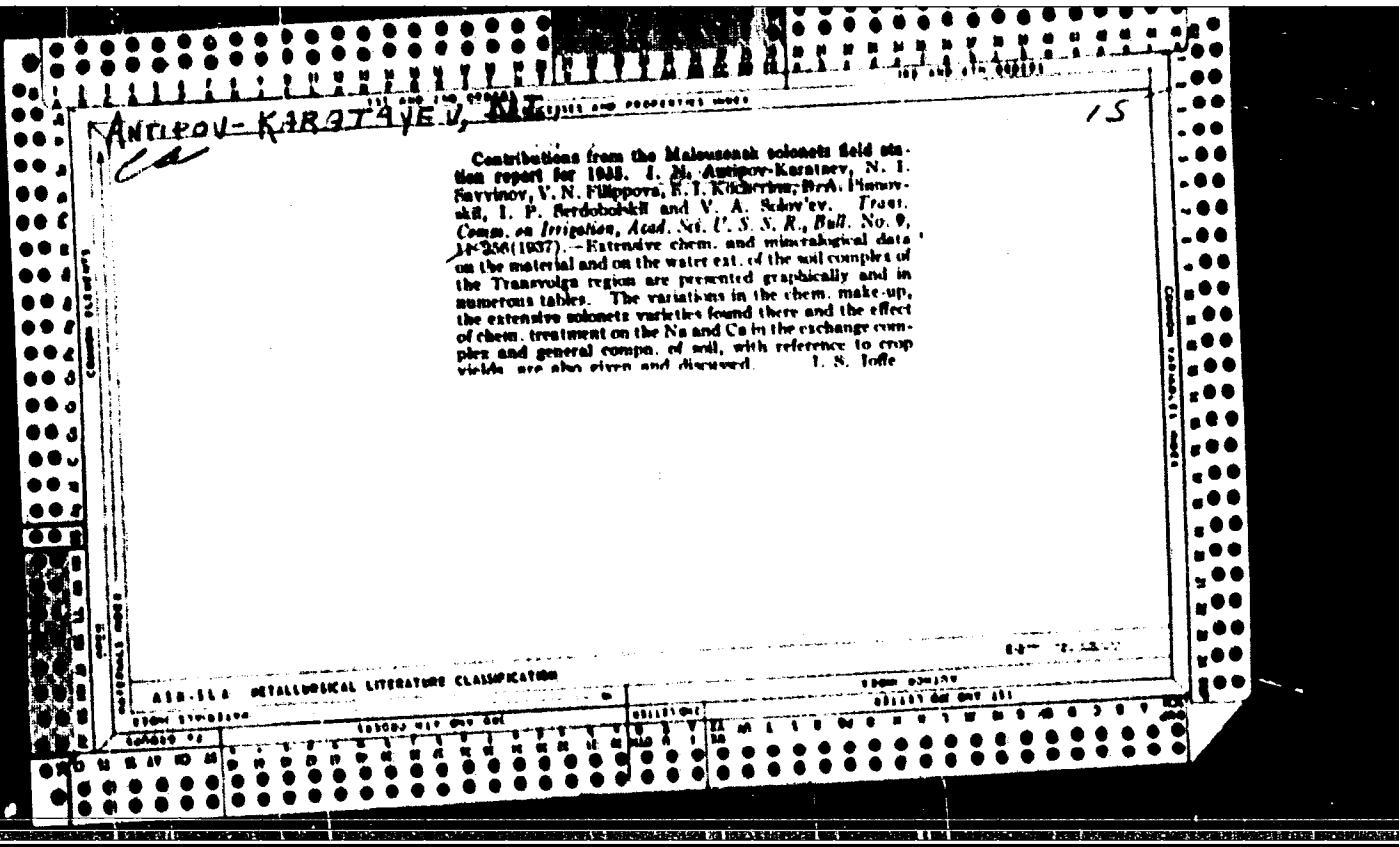
S-07-A-10-10-10-10

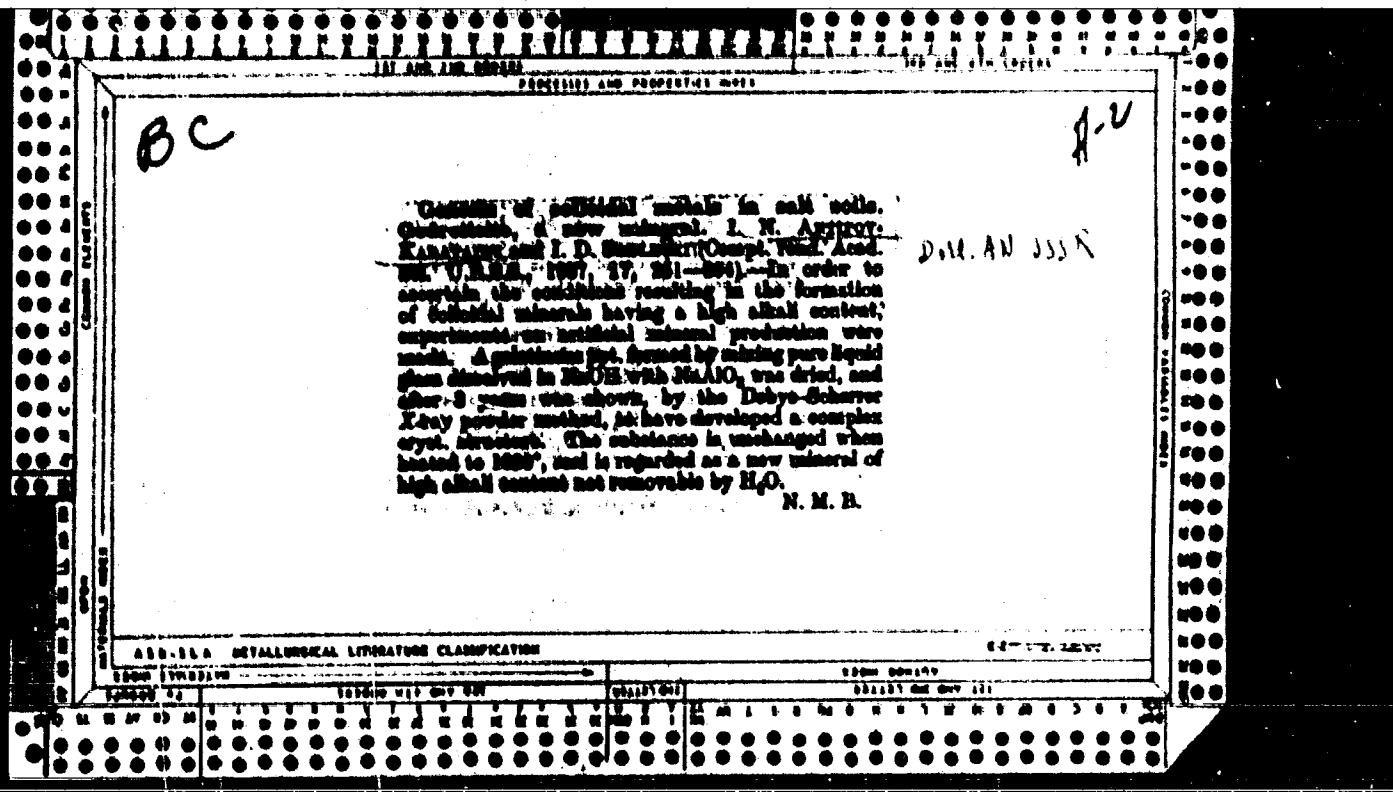
SHELF NUMBER

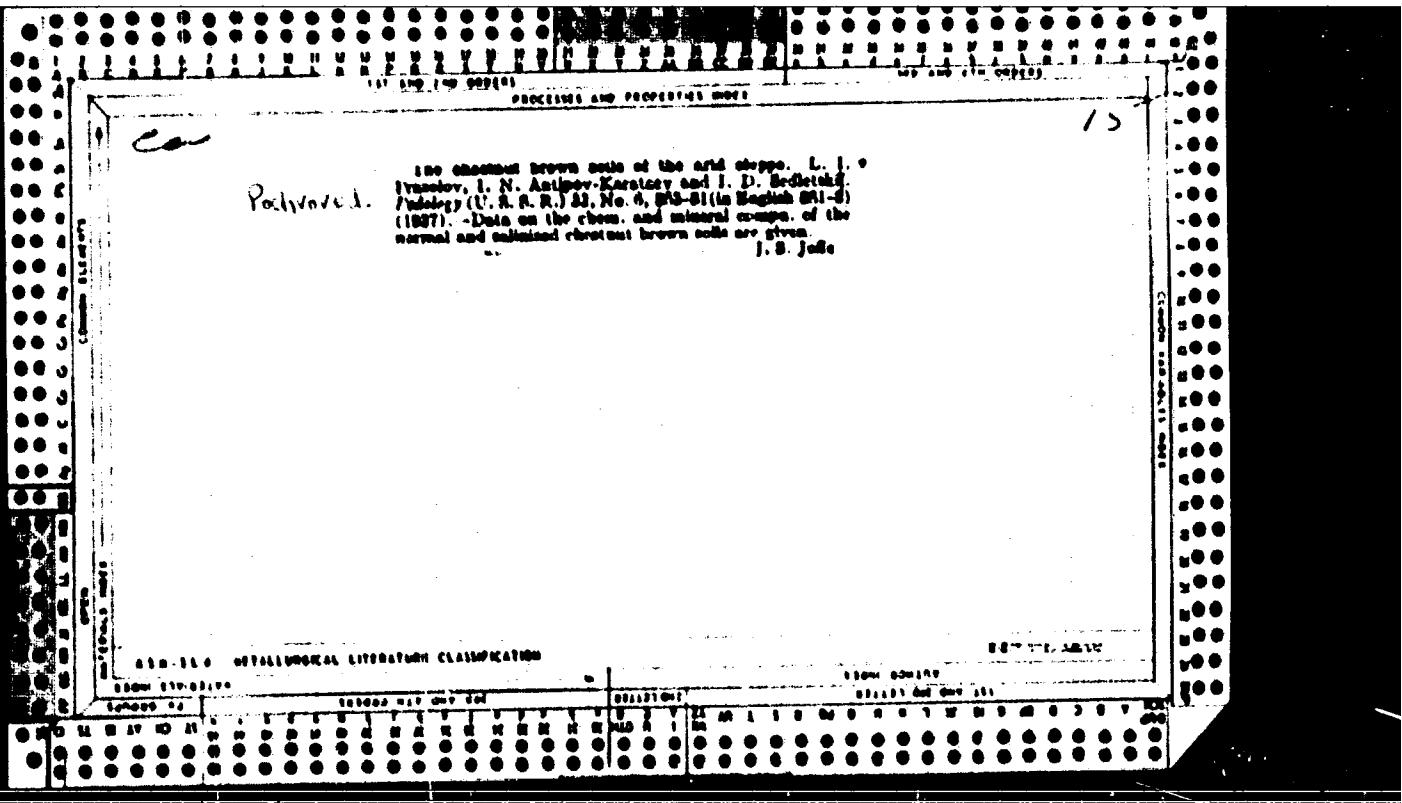
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The physicochemical processes of solonetz formation  
 I. N. Antipov-Karyagin and I. D. Sedetskii. *Podol'g*  
 (U.S.S.R.) 11, No. 6, 582-607 (1937).—Extensive data are  
 presented on the genesis of a no. of soil profiles and of the  
 colloid fraction. Total analyses, base-exchange capacity  
 and type of cations, x-ray analyses of the colloid fraction,  
 and mineral analyses of these are given. The process of  
 solonetz formation is extensively reviewed and discussed.  
 It is pointed out that the increase of Mg in the exchange  
 complex is not the cause of the solonetz properties, but  
 the result. Soils containing Na in quantities less than 8%  
 of the total absorbed bases are not solonetic; from 8-10%  
 Na they are weakly solonetic; from 10-15% Na they are  
 solonetic soils; and from 20% and more the soils are  
 genuine solonets. J. B. Jode

15

**APPROVED FOR RELEASE: 06/05/2000**

CIA-RDP86-00513R000101720009-1"

The chestnut soils. L. I. Pranikov and I. N. Antipov.  
Karpinsk. Pochvy S. S. S. R., Akad. Nauk S. S. R.,  
Karpinsk. Izd. in-ta Dohodchogo i. No. 196 (1959); Akad.  
Referat. Zhar. 1959, No. 10, 88-90.—In the chestnut soils  
the humus horizons 0/18-30-40 cm. thick and of a dark  
chestnut color; the soils are low in humus, and weaker in  
macrostructure than the chernozem soils with a total  
structure coeff. of 64-74%; there is a considerable dif-  
ferentiation of the soil profile; owing to interaction of  
 $\text{CO}_2$  with the hydrolysis products, primary silicates and  
carbonates of Ca and Mg accumulate. Absorption capac-  
ity is relatively high; satn. with Ca is 0.3-70% and with  
Mg 30-35% of capacity; content of absorbed Na is less  
than 5%;  $\text{NaO}_2\text{H}_2\text{O}$  is approx. 3.0. Montmorillonite  
and smectite predominate in the clay minerals; total  
 $\text{Fe}^{2+}$  is 0.06-0.18%, total K 1.2-2.3%, N 0.1-0.2%  
(C/N = 10-13), NH<sub>4</sub> 15-60 mg./kg. The formation  
of the dark chestnut soils is regarded as a stage preceding  
the formation of the chernozem soils. This stage is char-  
acterized by the removal of the water-sol. salts with their  
accumulation in the lower part of the profile, the begin-  
ning of the hydrolysis of the primary minerals with the  
formation of alkali and alk. earth carbonates, a consider-  
able humification with a simultaneous decompos. of the  
org. substances which cause the formation of the alkali  
and alk. earth carbonates accumulating in the limits of  
the soil profile and in the upper layers of the mother rocks.  
W. R. Henn

CA

15

Pechersk

The velocity of equalization of the composition of exchange cations during the mixing of various horizons of soils. (Study of the theory and practice for the production of rich tillable layers of soils) I. N. Antipov-Karataev and V. N. Filippova. *Zemledelstvo S.R.* 1939, No. 2, p. 12-9; Khim. Kifoz. 1947, No. 11, 49. Several samples of soils and clays were sated with various kinds of cations. Thin layers of these samples (which differed in their content of exchange cations) were laid from each other with sulphuric acid with water on the upper layer level and kept for 2 and 6 months. After these periods the content of the absorbed cations in the various layers was determined. The exchange Na was completely redistributed between the soils and clays during 2 months. The redistribution of the bivalent cations (Ca and Mg) was also practically complete after 2 months. These experiments were performed to solve the problem of increasing the tillable layer of the soil. A deep plowing of the podzolic soils causes a mixing of the eluvial and illuvial horizons of the soil and, owing to the redistribution of the cations between the horizons, the degree of saturation increases. In such cases smaller doses of lime are required. In the absence of the redistribution of absorbed Na, causes a decrease of the adsorptive properties; thus smaller addition of gypsum are required. W. R. Henn

AIA-SEA DETAILED LITERATURE CLASSIFICATION

PROCESSES AND PROPERTIES OF  
Saline marsh soils, solonetz soils and the improvement  
of solonetz soils. IV. N. Antipov-Karataev. Zn. Ural.  
Annyi Izdatel 1959, No. 8; 32-40. Relyazh. Republ. Zhet.  
1959, No. 12, 41.-Agrotech. methods are given for the  
addn. of gypsum to saline marsh and solonetz soils under  
conditions of irrigated and unirrigated agriculture.

W. R. Henn

650-104 METALLURGICAL LITERATURE CLASSIFICATION

Cation exchanges in soils and in solutions of mixtures of neutral salts. I. Anionic heteropolymer and anionic organic humic acids. *J. Soil Sci.* 3, 419-317 (1951). Chinese & Industrial 43, 254. A study of cation exchanges in chitosan soils for the systems Ca-Mg, Na-Ca, Ca-Mg + Na and Mg-Ca + Na shows that the corresponding adsorption reactions obey the law of mass action and that the exchanges are perfectly reversible and take place under conditions of strict equivalence.

Koll. Zhou.

CA

The genesis of the illuvial horizons of solonets. I  
A. Antunes Kacelacy. Pedology 1975, 5, 8-10. (Received)  
No. 1. Silty loam English soil. A distinction is made by  
between loampan and clay pan, the former being cemented by  
silicate and carbonate pipes; the clay pans are the result  
of the dispersion of the colloids. T. S. Lake

The laws governing the interaction of clays and soils with electrolytes. I. N. Anupya, Ratanlal S. Patel and V. A. Chavare. 1976. Noida, India. East. J. S. R. 120, 5. (from English translation). The mass law equation was applied in the study of ion exchange in soils and clays by keeping const. the ion content of the soils and clays. A linear relation between the pH and the electrolyte. A linear relation between the pH and the magnitude of ion exchange was established. Extensive amounts of data and graphs are presented on the effect of the exchangeable cations, H, Ca<sup>2+</sup>, K<sup>+</sup>, Na<sup>+</sup> and statuary of soil and clay minerals. The mutual relationships of the clay, adsorption, infiltration, suction force and drying properties as affected by electrolytes were extensively investigated. (E.S.)

Vest. Dokuchayev  
Pechovsk Inst.

CA

14

Chemical improvement of solonetz soils in U. S. S. R.  
I. N. Antipov-Karataev. *Chernozem. Novosibirsk. Agr.*  
*1940, No. 1, 22-31; Khim. Referat. Zhur.*  
*1940, No. 6, 61-2; cf. C. A. 35, 3781; 36, 6041.*--Addition  
of gypsum to the nonirrigated solonetz soils produces  
satisfactory results if the annual atm. precip. is approx. 300  
mm. or higher. Under irrigation, the effect of chem.  
treatment is obtained after 3-4 years. Data on crop  
yields of various plants and changes in the contents of  
exchange Na are given.  
W. R. Henn

Ca

## PRINCIPLES AND PRACTICE OF SOIL

15

## Determination of exchange constants for the soil cations.

T. N. Antipova-Karntseva and I. N. Antipov-Karntsev.

Pedology (U.S.S.R.) 1960, No. 2, 52-65 (in German, 61);

U. S. A. 34, 4218. - The quant. law governing adsorp-

tion of cations by soils are still to be established. This is

especially true of mixts of 2 or more salts. The theoretical

interpretation is given of the expnl. material published

earlier. In these expts. the exchange reactions between 2

and 3 cations in black soil were investigated. The condi-

tion of constancy of the ionic force of the solutions ( $J$ ) in saline and solonetz soils. The  $K$  values can be detdwas strictly adhered to in all expts. of a given series. The from 3-4 "points" if  $J$  is const.If 3 cations are present, value of  $J = \text{const}$ , was always calc'd. from the Lewis only two  $K$  values need be expnl.

The exchange of any other cation can be studied in presence of a known pair,

then reactions were shown to obey the law of mass action, if the concn. of the former in the given soil is known. With

They were also found to be reversible. For pairs of cat., the add of the above expts. it was possible to establish the

ions it was found:  $A_1/A_2 = G_1/G_2$ , where  $G_1$  and  $G_2$  exact ratios Na:Ca and Na:Ca + Mg<sup>2+</sup> in the soil solns.are the concn. of cations in equiv. solns.;  $A_1, A_2$  are the at which the Na ions begin to penetrate into the colloidal

compx. of the adsorbed cations for the case where both of complex of the soil (the potential saltiness of soil). Such

them are of the same valence. When valences are differ- quant. expts. are relatively more simple than the difficult

ent:  $A_1/A_2 = G_1/G_2$ , for the case where one is uni- empirical procedures used in the present-day practice of

valent and the other is bivalent. When 3 ions are present, saline and amelioration. The next problem is to establish

the value of  $K$  between them taken in pairs will be the same accurately the amplitude of fluctuations in the above. A

within 10-15% as if the 3rd cation were absent. Hence values for various cations in different soils. Then it will

the exchange const.  $K$  is practically independent of the be possible to calc. beforehand, on the basis of the compo-

nature of ions other than the given pair, and of the value of soil solns. and of the chem. nature and concn. of each

 $J$  of the adms. The av. values found are for black soil: cation present in the soil, the best measures to be followed $A_{\text{Na}} = 0.380$ ;  $A_{\text{K}} = 0.000$ ;  $A_{\text{Mg}} = 0.111$  in practical and theoretical problems. 31 references.Each one of these can be calc'd. from the 2 others:  $A_{\text{Ca}}$ .

C. S. Shapiro

## ASL-154 METALLURGICAL LITERATURE CLASSIFICATION

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100
1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100

CF

KvN-2405  
Application of different methods of separation of clay for investigation of the mineralogical composition of its colloidal fractions. I. N. Antipov-Karatsev, L. V. Sushanskii and S. M. Yusupova. Colloid J. (U.S.S.R.) 6, 133-44 (1960). -- The international method is compared with the method of fractional precipitation (cf. Tsvet, p. 1, 31, 1950). The second method could be used for the spm. of amorphous and semicryst. samples from the crust. The 1st method need the mixt. of minerals into groups of 2-3 minerals and, therefore, guaranteed completeness and accuracy of their data (the no. of lines on the contourgram never exceeded 10-15). A. A. P.

APPENDIX METALLURGICAL LITERATURE CLASSIFICATION

"APPROVED FOR RELEASE: 06/05/2000

CIA-RDP86-00513R000101720009-1

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ca

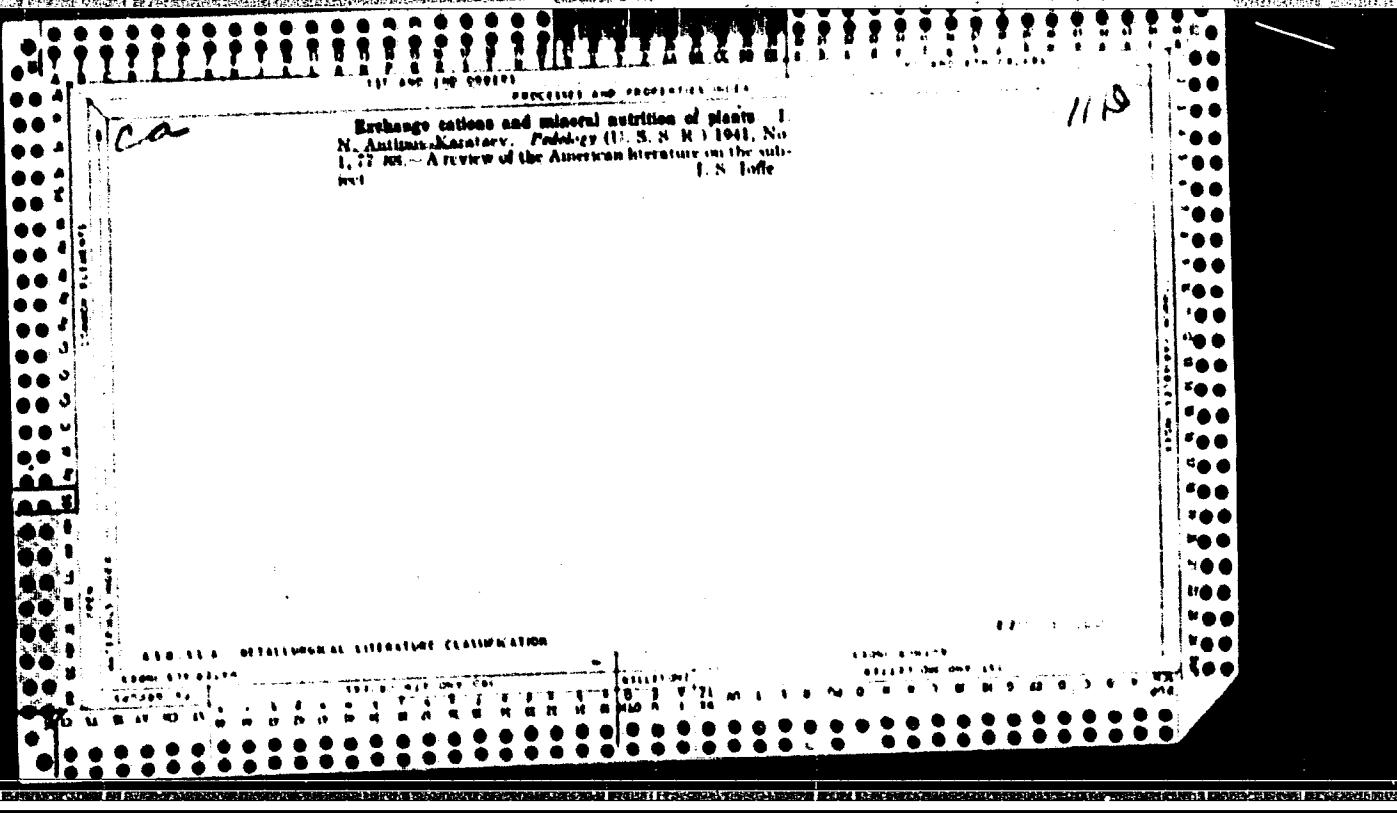
Recent work on soil colloids. T. N. Anupam, Karan Singh  
Published in Indian J. Soil Sci., No. 1, 1960, C. S. Shapiro

APPROVED FOR RELEASE: 06/05/2000

CIA-RDP86-00513R000101720009-1"

"APPROVED FOR RELEASE: 06/05/2000

CIA-RDP86-00513R000101720009-1



APPROVED FOR RELEASE: 06/05/2000

CIA-RDP86-00513R000101720009-1"

CA

The culture of sugar beets on soils of the solonetz complex. I. N. Antipov-Karataev. *Aridology* (U. S. S. R.), No. 3, 26-36 in English, 33 (1942). Data are presented on the capacity of chloride-sulfate and soda types of saline soils. It is pointed out that these soils can be improved for sugar-beet culture by deep plowing and incorporation of available org. matter. As a war emergency this type of improvement, without the aid of chem. agents, is possible. J. B. Joffe

CA

V2

The genesis of the colloidally dispersed soil minerals.  
I. N. Antipov-Karatav, G. M. Ponomarev and I. D.  
Sedeltseff. Pedology (U. S. S. R.) 1942, No. 3-4,  
34-9 (in English, 39-60). Using the x-ray method, the  
authors detd. the types of minerals found in the fraction  
less than 0.2  $\mu$ . In the profile of a light-gray earth (gray  
soil of regosol) on porphyry granite, to a depth of 60  
cm, the following minerals were found; beidellite and  
muscovite-sercite. In a light-chestnut brown soil on  
quartz porphyry, to a depth of 20 cm, the following were  
found; montmorillonite and muscovite-sercite. In a  
light-chestnut brown on granite to a depth of 60 cm, the  
same minerals were found. In medium-rich humus  
chernozem on granite eluvium montmorillonite only was  
found in the first 10 cm. From this depth to 80 cm, the  
montmorillonite and muscovite-sercite were found. In  
degraded chernozem on porphyry-schist, to a depth of  
20 cm, the montmorillonite and muscovite-sercite minerals  
were found. J. S. Joffe

CA

15

Methods of investigating the absorptive power of soils  
in line with the decisions of the International Congress of  
Soil Science in Oxford, 1935). I. N. Antipov-Karatsev  
and R. S. Firsov. *Pedology* (U. S. S. R.) 1942, No.  
3, p. 101 (orig. English, 1932). A report on the basic ad-  
sorbed, exchange capacity, and unsat. of 7 soil samples  
distributed by the ministry to a no. of labs. The authors  
detd. the exchange capacity of the soils by electrolyzing  
and then treating them with a soln. of  $\text{CaCl}_2 + \text{Ca(OH)}_2$ ,  
of an ionic concn. = 0.05 (satd.  $\text{CaO}$  soln.). The quantity  
of Ca adsorbed was detd. potentiometrically. J. S. J.

ANTIPOV-KARATAYEV, Ivan Nikolayevich

"On the Method of Determining the Absorption Capacity and the Structure of Exchanging  
Bases in the Carbonate Soils," Pochvoved., Nos. 9-10, 1942.

The soil as a polydisperse system as developed in the U. S. S. R. for 25 years (1917-1942). I. N. Antipov-Kuratcov, *Pedology* (U. S. S. R.) 1943, No. 8, 3 (in English, 20).—A. presents and analyzes critically the following angles of the subject: (1) the solid phase of the soil as a hetero- and polydispersed mass; (2) a total chemical analysis of the solid phase of the soil and its fundamental granulometric fractions; (3) a complete mineralogical analysis of the fundamental granulometric fractions of the soil, especially its highly dispersed fractions; (4) the solid phase of the soil and soil salts, their interaction in the light of: (a) electrometric phenomena and amphoteric behavior of soils; (b) the solid phase of the soil and water; (c) absorptive capacity of soils in relation to electrolytes and nonelectrolytes with the resulting acidity and alkalinity of soils; (5) the laws governing the dispersion of the soil matrix: coagulation, precipitation, structure and thixotropy; (6) the solid and gaseous phases and their interaction; (7) the application of the studies of the soil as a polydispersed system to soil mineralization, high-speed construction, ceramics and war effort. J. S. J.

ANTIFOV-KARATAY'V, I. N.

"Changes observed in the Dispersity of Soil Masses and Problems of their Practical Utilization," Pochvoved., No. 7, 1943

ICA

15

Methods of determining the absorptive capacity and composition of exchangeable ions in carbonate soils  
I. N. Antipov-Karpinsky and N. I. Zaitseva. Pedology  
(U.S.S.R.) 1943, No. 3-10, 42 (English summary)

The method adopted consists of taking 4 g. of soil and  
with 10 (0.5 N)  $\text{BaCl}_2$  and 0.01 N  $\text{Ba}\text{C}_2\text{O}_4$  (or acetate are used) and  
treating it with 30 cc. 0.05 N  $\text{K}_2\text{CO}_3$ , free from  $\text{KHCO}_3$ .  
The mixt. is allowed to stand for 2 to 4 hrs. and shaken  
occasionally. The suspension is allowed to settle for 40-  
48 hrs., or a filtrate obtained by passing it through a col-  
loidal filter. To 30 cc. of the filtrate 10 cc. of 0.1 N  $\text{Ba}\text{S}_2\text{O}_3$   
is added. The  $\text{CO}_2$  formed is driven off by heating  
and the residual filtrate titrated with  $\text{NaOH}$  (0.05 N) with  
a mixt. of bromine cresol purple and bromothymol blue as  
the indicator. All soils gave good results, except for the  
thermous or any other soil sample rich in org. matter.  
For these soils an ab. amt. of  $\text{MgSO}_4$  (Sushko, C.A. 30,  
4259) is suggested.

*Ca*

Composition of cation dispersion minerals and the  
metathesis capacity of glauconites from Saratov. I. N.  
Antipov-Kurnitsky and L. D. Nedel'skii. *Comp. rend.*  
*USSR Akad. Nauk SSSR*, 116-17(1943). — X-ray patterns  
of glauconites before and after saturation by solutions of chlorides  
of Li, Li, Mg, Na and Ba showed no structural changes.  
These samples are found to consist of 60% glauconite, 4%  
montmorillonite and 10% berkeite. At pH = 9, the ex-  
change capacity of one glauconite is 31.08 milliequiv. of  
 $M\text{gCl}_2$  per 100 g. material; a change of 1 pH unit  
changed its capacity by 4.0 milliequiv. The correspond-  
ing figures for permutite are 430 and 49 milliequivs.

Cyrus Fehlman

*Date 10/30/87*

ANTON'YAKA-SATAYEV, I. N.

"The Struggle against Salting of Irrigated Soils," Pochvoved., No. 2, 1945

31

B

3351 Rekomendatsii Polosykh i Laboratoriyskikh Issledovaniy Pechat. Tom IV. Sovremennoye Metoda Issledovaniya Fiziko-Khimicheskikh Svoystv Pechat. (Handbook for Field and Laboratory Investigations of Soils Vol. IV. Contemporary Methods of Investigating the Physico-Chemical Properties of Soils). Nos. 1-2. L.N. Antipov Katalay, editor. 162 and 291 pages. 1943 and 1947. Moscow and Leningrad, USSR (SFSR) (Abstr.)

First volume (no. 1) contains 3 separate articles on methods for study of soils. Second volume (no. 2) likewise contains 7 articles on specific soil analysis and investigation methods. Graphs, tables, diagrams, and illustrations.

CA

The progress made and the problems of the chemistry of  
soils in the U.S.S.R. In connection with 30 years of the  
Soviet regime. I. N. Antipov-Karatary. Pedology (U.  
S.S.R.) 1947, no. 6. A review of the literature on the  
subjects of geochemistry, phys. chemistry, and chemistry  
of minerals in relation to soil problems. I. N. I.

CA

The movement of copper in soils. I. N. Antipov-Karginov. Pedology (U.S.S.R.) 1947, 6(3-4). Cu is retained by soils and clays in two forms: exchangeable and chemisorbed (not exchangeable). Montmorillonite clays and soils high in org. matter tie up the Cu much more tightly than the t's and s's. In the case of smectitic-clay soils, the stability of adsorption is reversed. It is suggested to use D.B.N. 1146 for the estimation of exchangeable Cu. The entire subject of adsorption and immobilization of Cu is reviewed, and a list of 20 references given.

100-104 METALLURGICAL LITERATURE CLASSIFICATION

CA

AA

The brown forest and dark brown forest soils. I. N.  
Antipov-Karginov. Pedology (U.S.S.R.) 1947, no. 7: 713  
On the basis of analyses of the forest soils in Ukraine  
and their coloids, specifically the North Alak's and North  
Ketka. A. concludes that these soils have to be regis-  
tered as an independent genetic type. J. B. Inde

**CA****Nature of the uptake of ions by clays and soils III****Uptake of univalent and bivalent cations of Chernozem soil and humic acid**I. N. Antipov-Karatayev, G. M. Kader, and V. N. Filippov. *Zhur. 9*, 315 (1971).

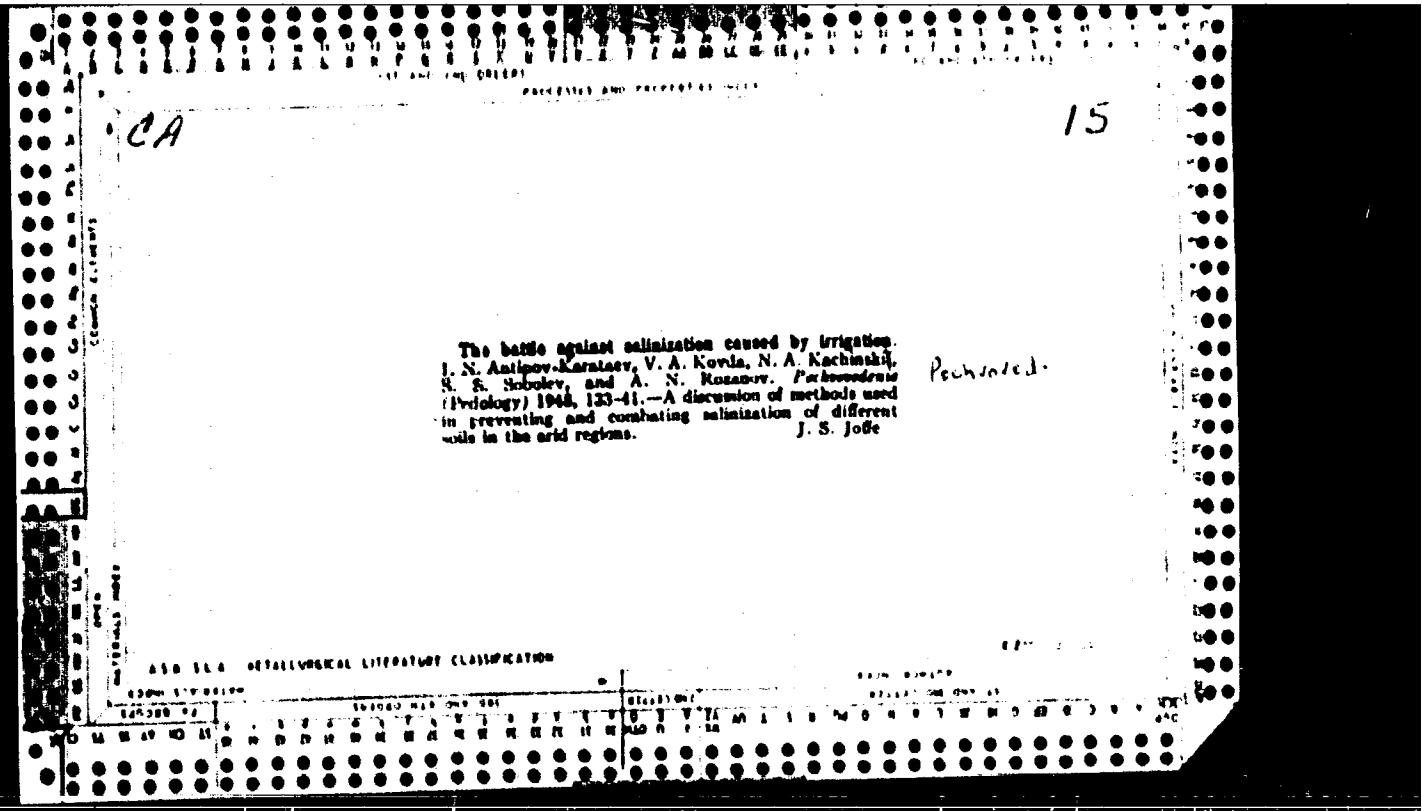
Solvent with cation  $K_1$  is shaken with a soln. of the chlorides of  $K_1$  and  $K_2$  in which the ratio  $K_1/K_2$  is varied between 1.0 and 0.1. If the total contents of  $K_1$  and  $K_2$  in the soln. are  $G_1$  and  $G_2$ , and the units in the solns.  $X_1$  and  $X_2$ , then  $X_1G_1/X_2G_2 = k$  is the "equilibrium const." The uptake was reversible but  $k$  varied with  $K_1/K_2$  for Ca/Mg or a Ca humate from peat, for K/Na and Ca/Pb on Chernozem, and also for Ca/Cu on Chernozem when the exchange occurred in an alk. soln. The uptake was irreversible for Cu/Hg, Ca/Cu in HCl, and Ca/Al (at pH 4.3) on Chernozem, i.e., Hg, Cu, and Al could not be displaced by Ca. The ion strength in all expts. was 0.0225 or 0.0250. The differences between humic acid, Chernozem, and montmorillonite are discussed.

**IV. Uptake of bivalent cations by red brown clay and some soils***Zhur. 10*, 73 (1972).

The uptake was reversible and the "equal const."  $k$  was independent of  $K_1/K_2$  for Ca/Mg on a redbrown clay (I) (whose satn. capacity  $m = 35$  milliequiv. per 100 g.), a red soil ( $m = 22$ ), and a peat soil ( $m = 12$ ),  $k$  being 1.5, 1.1, and 1.1, resp. The Cl<sup>-</sup> concn. was 0.015 N in all cases. The uptake was reversible but  $k$  varied with  $K_1/K_2$  for Ca/Pb and Ca/Hg, and the reversibility was poor for Ca/Cu in both aqu. and alc. solns. (alton). The greater the value of  $k$  the greater is the selectivity of a soln. Bentonite clay is less selective than soils (contg. K, Al, and SiO<sub>2</sub>). The uptake decreases in the series Pb > Fe > Ca > Mg for Chernozem and Hg > Cu > Fe > Mg for montmorillonite clay. The variation of  $k$  with the ratio  $K_1/K_2$  is a measure of heterogeneity of the adsorbent. Kolin and humic acid are particularly heterogeneous. The irreversible uptake of Hg by Chernozem is greater than by other soils, presumably because of the org. matter in Chernozem.

**V. Uptake of heavy metal ions by clay and soils in dynamic conditions**I. N. Antipov-Karatayev, M. A. Pavlik-Khlebina, M. S. Merkulova, and V. I. Garben-shchikova. *Zhur. 10*, 5. Solns. of HgCl<sub>2</sub>, PbCl<sub>2</sub>, and CuCl<sub>2</sub> were filtered through a column 5 cm. wide of argilaceous soil, satd. with 10 mg. equiv. Ca per 100 g. at a rate of 1 cm. min.<sup>-1</sup>. The heavy metal ions broke through after passage of 1.1 of soln. I = 177, i.e.,  $H$  is the height of the column (10.6 cm.). For 10 mg. equiv. HgCl<sub>2</sub> and 0.16 N PbCl<sub>2</sub>, the const.  $k$  was 277 and 100.1, cm., resp. The const.  $k$  was almost identical for HgCl<sub>2</sub> and PbCl<sub>2</sub>. CuCl<sub>2</sub> behaved like PbCl<sub>2</sub>. Before the break through, the soil behaved like PbCl<sub>2</sub>. Before the break through, the soil behaved like PbCl<sub>2</sub>. Before the break through, the soil behaved like PbCl<sub>2</sub>. Before the break through, the soil behaved like PbCl<sub>2</sub>. Before the break through, the soil behaved like PbCl<sub>2</sub>. Before the break through, the soil behaved like PbCl<sub>2</sub>.

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USSR/Soil Science  
Geography

Feb 48

"A Trip Into Bulgaria," I. P. Gerasimov, I. N. Antipov-Karatayev, 2½ pp

"Pochvoved" No 2

Describes desert, forest soils, soils of high altitude peat regions, and azonal or nonclimatic soils. Genetic types of soil in Bulgaria. Trip was of short duration; nevertheless, authors were able to obtain fair amount of data.

PA 62T101

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USSR/Soil Science  
Ions  
Absorption

Mar/Apr 1948

"The Nature of the Absorption of Ions by Clays and Soils: IV. Absorption of Bivalent Cations by Red-Borax Clays and Certain Soils," I. N. Antipov-Karatayev, G. M. Kader, V. N. Filippova, Soil Inst, Acad Sci USSR, Moscow, 10 pp.

"Kolloid Zhur" Vol X, No 2

Describes replacement absorption of magnesium ions in Ca-clays, Ca-sub-ash soils and Ca-red clays, absorption of heavy metal ions on Ca-clay-covered surfaces, and results of the observations. Submitted 22 Mar 1947.

707107

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Nov/Dec 48

"The Nature of Ion Absorption by Clays and Soils:  
V. Absorption of Heavy Metal Ions by Clays and  
Soils Under Dynamic Conditions," I. N. Antipov-  
Karataev, M. A. Pasvik-Khlopina, M. S. Merkulova,  
V. I. Grebennichikova, Soil Institute, and Radium  
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Experiments show that the method of sorption  
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USSR/Chemistry - Ions (Contd)

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5. Institut geografii Akademii nauk SSSR (for Armand, Fedorovich, and Poteев).
6. Geograficheskiy fakul'tet Moskovskogo gosudarstvennogo universiteta (for Zubov and Alisov).
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